

The microbial carbon pump and climate change

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Abstract

The ocean has been a regulator of climate change throughout Earth's history. One key mechanism is the mediation of the carbon reservoir by refractory dissolved organic carbon (RDOC), which can either be stored in the water column for centuries or released back into the atmosphere as carbon dioxide depending on the conditions. The RDOC is produced through a myriad of microbial metabolic and ecological processes known as the microbial carbon pump (MCP). Here, we review recent research advances in processes related to the MCP, including the distribution patterns and molecular composition of RDOC, links between the complexity of RDOC compounds and microbial diversity, MCP-driven carbon cycles across time and space, and responses of the MCP to a changing climate. We identify knowledge gaps and future research directions in the role of the MCP, particularly as a key component in integrated approaches combining the mechanisms of the biological and abiotic carbon pumps for ocean negative carbon emissions.

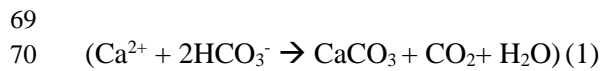
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In this Review, Jiao, Robinson and colleagues examine recent advances related to the microbial carbon pump, exploring its role in the carbon cycle and climate change, and proposing future research directions and approaches to ocean negative carbon emissions.

[H1] Introduction

The ocean represents the largest active carbon reservoir on Earth harbouring 50 times more carbon than the atmosphere and therefore plays a critical role in regulating climate change^{1,2}. The major components of the ocean carbon reservoir include dissolved inorganic carbon (DIC), dissolved organic carbon (DOC), particulate organic carbon (POC), and particulate inorganic carbon (PIC). These carbon pools are mediated by physical, chemical, biological and biogeochemical mechanisms known as the solubility carbon pump (SCP), microbial

58 carbon pump (MCP)³, biological carbon pump (BCP)^{4,5}, and carbonate counter pump (CCP)⁶ (FIG. 1). The
59 SCP depends on abiotic controls of the partial pressure difference between the atmosphere and the ocean
60 surface, and is the mechanism by which increasing atmospheric carbon dioxide concentrations lead to ocean
61 acidification⁷. The MCP depends on the microbially mediated physiological and ecological processes
62 transforming DOC from labile (LDOC) to refractory (RDOC) states. The latter can escape further biological
63 and chemical oxidation, and be sequestered in the water column for decades, centuries, or millennia³. However,
64 the turnover of RDOC is under debate⁸. The BCP is based on carbon fixation by phytoplankton in the surface
65 ocean and the subsequent export of organic carbon to the ocean interior and seabed. The BCP has been studied
66 for about half a century⁹⁻¹²; therefore, only the part that is intertwined with the MCP will be addressed in this
67 review. The CCP is based on the precipitation of carbonate by calcifying organisms which release carbon
68 dioxide (CO₂):



71
72 and thus, it is called a counter pump. The proportion of the produced CO₂ that is released to the atmosphere
73 depends on the chemical, biological, and physical conditions of the water, and thus needs to be addressed in
74 the context of an ecosystem and from the perspective of interactions between the MCP, BCP, CCP, and SCP.
75 Knowledge of the interactions between the MCP, BCP, CCP, and SCP is important for a better understanding
76 of the controlling mechanisms of carbon sequestration in the forms of POC, PIC, and RDOC in the ocean.
77 RDOC is an enigmatic part of the ocean carbon reservoir and plays a unique role in the regulation of carbon
78 cycling and climate change from a long-term dynamic perspective. It is often less appreciated that the delicate
79 balance between ocean total DIC and total alkalinity (TA) can be easily shifted by seemingly small changes
80 mediated by the MCP, and this has been of critical importance in regulating climate change over the millennia
81 of Earth's history¹³. If only 0.5% of DIC is converted to RDOC, it could result in a 25 parts per million (ppm)
82 decrease in the partial pressure of CO₂ (*p*CO₂) at the sea surface and in the atmosphere, which is about 28% of
83 the CO₂ change between glacial and interglacial times, or equivalent to 12-years of cumulative *p*CO₂ change
84 due to anthropogenic carbon release around the world today^{13,14}. In this context, assessment of the MCP is key
85 in determining the production and fate of marine RDOC. In this Review, we explore the characterization of
86 RDOC compounds, their recalcitrance and interactions with microbial diversity, factors influencing their
87 production and distribution, their role in paleoclimate, and their response to climate change¹⁵⁻¹⁸. We also define
88 future research directions needed to fill crucial knowledge gaps and to assess the potential synergistic effects
89 of the combined BCP–CCP–MCP–SCP¹⁹ to increase the removal of atmospheric carbon dioxide to achieve
90 ocean negative carbon emissions (ONCE) and thereby mitigate the impacts of climate change.

91
92 **[H1] The microbial carbon pump**

93 The oceans are full of invisible but globally impactful microorganisms²⁰. They are the main bearers of biomass
94 and mediators of energy flux in the marine ecosystem, assuming a pivotal role in biogeochemical cycles.
95 Despite substantial advances in our understanding of the functioning of marine microorganisms, several
96 enigmas remain. DOC accounts for more than 90% of the total marine organic carbon pool²¹ and serves as the
97 energy and carbon source of heterotrophic microorganisms supporting the microbial loop²². Thus, how can a
98 huge reservoir of DOC remain despite the myriad of hungry microorganisms? This situation has puzzled
99 microbiologists long before the concept of the MCP³ emerged a decade ago. The MCP refers to the microbial
100 transformation of organic carbon from labile to refractory states, that is, RDOC that resists further utilization,
101 although the turnover of RDOC components is still not known. RDOC can remain in the water for a long time,
102 leading to accumulation and maintenance of the large marine DOC pool to approximately 642 petagrams (Pg)
103 of carbon²³, almost equivalent to the total carbon inventory of atmospheric CO₂.

104 A critical step in moving from a conceptual framework to a recognized theory is through experimental studies
105 to prove that RDOC is produced via the MCP. One piece of evidence came from the Aquatron mesocosm
106 experimental study performed at Dalhousie University in Canada using a tower tank containing around 117,100
107 litres of water. In the experiment, RDOC with a fingerprint (as determined by Fourier Transform Ion Cyclotron
108 Resonance Mass Spectrometry (FT-ICR-MS)) similar to that of oceanic water from 2,000 metres was produced
109 from labile DOC during the incubation²⁴. Several key microbial pathways related to the production of RDOC
110 were identified via metagenomic analysis in the Aquatron experiment²⁵. Even if the proportion of RDOC
111 production to LDOC consumed is tiny, RDOC can accumulate over time and become a large carbon pool. It
112 remains however challenging to untangle the internal cycling because no direct evidence reported to date that
113 the DOC pool in the deep ocean is increasing but fluctuations of RDOC pool must exist in Earth's history^{26,27}.

114 The MCP involves heterotrophic, autotrophic, and viral activities, including decomposition and degradation,
115 synthesis and conversion, and lytic processes^{3,28,29}. These MCP processes are successive and dynamic under
116 changing environmental conditions. The products of the MCP, RDOC compounds, are differentiated into
117 RDOct and RDOcc fractions³⁰. RDOct is defined as the fraction that is intrinsically recalcitrant in a given
118 environmental context, while RDOcc is composed of diverse molecules with varying concentrations, but
119 which are below their respective thresholds for efficient uptake³⁰. The differentiation of RDOct and RDOcc
120 provoked a debate on ‘recalcitrance’ or the ‘dilution hypothesis’³¹⁻³⁴ which further led to the quantification of
121 RDOct and RDOcc. Eventually, RDOct was shown to be the major component of RDOC^{33,35}.
122 There are fundamental differences between the MCP and other carbon pumps. For example, the BCP occurs
123 through photosynthetic production of organic particles and their sedimentation from the surface to the seabed
124 while the MCP does not depend on such physical transport processes. The MCP theory suggests that high
125 levels of photoautotrophic carbon fixation do not necessarily correlate with high levels of carbon
126 sequestration³⁶. Thus, the most productive waters are often carbon sources to the atmosphere rather than
127 sinks^{37,38}. The MCP also changes the elemental composition of organic matter in the ocean, affecting the carbon
128 (C):nitrogen (N):phosphorus (P) ratio by depleting the DOM pool in N and P while retaining C (3511:202:1)
129 ^{3,39}, which is far from the Redfield ratio (106:16:1) for phytoplankton biomass. By releasing N and P while
130 preserving carbon, the MCP facilitates nutrient recycling for primary production and carbon sequestration, and
131 raises questions about the composition of RDOC, its long-term stability and its sources. It seems that deep
132 ocean carbon fixation by chemolithotrophs plays a substantial role in DOC production⁴⁰, but to the best of our
133 knowledge, no data to date demonstrate how much RDOC is produced by chemolithotrophs.

134

135 **[H1] The nature of refractory dissolved organic carbon**

136 RDOC was defined based on its bioavailability and turnover time^{3,41}, with the age of RDOC inferred from
137 radiocarbon dating of the bulk or broadly defined fractions⁴²⁻⁴⁴. However, deep ocean sources of Carbon-14-
138 free DOM have been found in recent years, suggesting potential biases in this dating method^{44,45}. These studies
139 challenged the conventional view that RDOC is evenly distributed in concentration and radiocarbon age
140 throughout the ocean. Instead, the data indicate that both the concentration and radiocarbon age of RDOC vary
141 depending on depth and location⁴⁶. The new evidence challenges our knowledge of deep ocean RDOC age,
142 and so assessment of the functioning of MCP may be a more promising route to determining-the production
143 and fate of RDOC.

144 Characterization of the composition of RDOC has improved substantially over the past decade with the
145 development of instrumentation and techniques^{23,24,47}. The composition of RDOC can be defined by its
146 molecular optical properties, which can be detected by ultraviolet-visible and fluorescence spectroscopy. The
147 marine fluorophores, which mimic the fluorescent signals of humic substances (FDOM_H) in freshwater
148 systems, were found in the products of marine autotrophic and heterotrophic organisms⁴⁸⁻⁵⁰. Given the
149 residence time of terrestrial FDOM_H and the constancy of marine FDOM_H in the deep ocean, microorganisms
150 contribute substantially to marine FDOM_H⁵¹⁻⁵³. FDOM_H is a proxy of RDOct as DOM fluorescence increases
151 with the microbial transformation of organic substrates^{51,52,54}.

152 The application of ultrahigh-resolution mass spectrometry in bulk DOM analyses unveiled the mystery of
153 unknown complex DOM compounds. A diverse group of carboxyl-rich alicyclic molecules (CRAMs) account
154 for a substantial fraction of the RDOC in the deep ocean⁵⁵⁻⁵⁸. They have been shown to accumulate during
155 microbial transformation processes, as evidenced by laboratory experiments⁵⁹, environmental surveys^{60,61},
156 incubation assays⁶² and metagenomic analyses²⁵. They are therefore another proxy of RDOC. However, the
157 limitations of the detection and extraction methodologies mean that CRAMs describe only part of the
158 complexity of RDOC. It is also important to note that not all CRAMs can be simply classified as RDOC^{63,64}.
159 A large fraction of the refractory compounds and molecules remains unknown even in the accessible data sets.
160 Dissolved black carbon (DBC) also forms part of the biologically recalcitrant components in the ocean⁶⁵⁻⁶⁷.
161 The major sources of DBC are land-based biomass burning and fossil fuel combustion, and this DBC is then
162 transported to the ocean via rivers and atmospheric deposition^{68,69}. However, recent evidence from isotope
163 studies suggests that DBC has biological sources⁷⁰. DBC is also derived from marine hydrothermal vents and
164 methane seeps^{71,72}. Given that the DBC pool in the global ocean is 9.5–14 Pg of carbon and that its carbon-14
165 age is about 23,000 years⁷³, its contribution to RDOC in terms of quantity and age appears insignificant
166 (estimated to be 1–2% and 4–8% respectively).

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168

169 **[H2] Microbial production and transformation of refractory dissolved organic carbon**

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171 The generation of RDOC by the MCP involves a series of sequential microbial processes³⁰. Different microbial
172 groups exhibit distinct capabilities in organic matter degradation. Certain microorganisms assimilate organic
173 matter released by other microbial taxa, while specific RDOC compounds are uniquely associated with
174 particular microorganisms^{74,75}.

175 Recent investigations into the key species participating in RDOC production and transformation underscore
176 the intricate mechanisms through which microbial activities convert labile DOM into a refractory state⁷⁵.

177 Enzymatic oxidation and hydrolysis are key processes in the microbial degradation of DOM. Studies on
178 enzyme diversity, the substrate specificity of enzymes (functional recalcitrance may arise from the requirement
179 for specialized enzymes), enzymatic kinetics (reaction rates and substrate affinity), and cellular allocation of
180 enzymes in kinetic metabolic networks⁷⁶⁻⁷⁹ have demonstrated the complexity and diversity of microbial
181 functions in transforming DOM, thereby influencing the composition and distribution of in situ DOM
182 composition. Several bacterial groups have been associated with the production of RDOC, such as members
183 of the *Flavobacteriaceae*, *Polaribacter*, *Roseobacter*, *Alteromonas*, and *Pseudomonas* families^{75,80}.

184 The addition of various DOM sources to bioassays, such as simple compounds or polymers⁸¹, phytoplankton-
185 derived DOM^{58,82}, soil or sediment DOM⁵⁴, led to the formation of chemically recalcitrant components (for
186 example, CRAMs). Picocyanobacteria, one of the major groups of primary producers in the global ocean,
187 produce typical deep ocean fluorescent DOM (FDOM_H) components from degradation products of
188 photosynthetic phycobilin pigments⁴⁸. The microbial degradation of carotenoids⁸³ is widely found in a variety
189 of algae and bacteria resulting in the production of CRAMs⁸⁴, and several bacterial groups such as the SAR202
190 clade, and the *Piscirickettsiaceae*, *Hyphomonadaceae*, and *Alcanivoracaceae* families have demonstrated
191 variable degradation rates for certain CRAM substrates⁷⁶⁻⁷⁹.

192
193 **[H2] Linking diverse microorganisms to complex refractory dissolved organic matter**

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195 The advent of advanced analytical tools such as Fourier Transform Ion Cyclotron Resonance Mass
196 Spectrometry (FT-ICR-MS) and nuclear magnetic resonance (NMR) after solid-phase extraction by modified
197 styrene-divinylbenzene polymer resins has greatly improved our understanding of the diversity and complexity
198 of DOM, both in terms of chemical composition and compound structure^{55,64}. At the same time, with the
199 development of omics approaches, the diversity of microbial communities has expanded from the taxonomic
200 level to genetic and metabolic functions. Recent studies have begun to address the challenge of linking the
201 interactions between chemical and microbial complexity (FIG. 2).

202 Microorganism–DOM molecular network analyses are important statistical tools for visualizing the complex
203 interactions between diverse microbial communities and DOM compounds^{85,86}. Network graphics not only
204 visualize the complexity of these interactions, but also provide a comprehensive picture of how the intricacy
205 varies temporally and spatially⁸⁶. The key microbial species involved in RDOC formation and distribution
206 patterns have been identified based on these analyses, along with the roles of specific microbial groups in the
207 formation and transformation of DOM, and their responses to environmental changes. For example,
208 Gammaproteobacteria, Betaproteobacteria, Cytophaga, Thaumarchaeota and Euryarchaeota were found to
209 contribute to the consumption and decomposition of phytoplankton-derived DOM^{74,87,88}.

210 Long-term dynamics observations at coastal time-series stations have revealed that fluctuations in the
211 environment and DOM are determining the structure of microbial communities⁸⁹. DOM components, such as
212 CRAMs, D-amino acids, highly unsaturated aromatic compounds, and nitrogen-rich compounds were
213 recognized by these correlation networks, indicating the linkage between the recalcitrance of DOM and
214 specific microbial species^{25,90}. Chemically recalcitrant compounds also frequently co-occurred with key
215 microbial species along latitudinal transects and other environmental gradients⁸⁵.

216
217 **[H1] Drivers of refractory dissolved organic carbon distribution in a changing ocean**

218
219 **[H2] Terrestrial versus indigenous dissolved organic carbon**

220 The MCP also exists in terrestrial soil environments, and is therefore referred to as the soil MCP¹⁵. Once the
221 soil RDOC is washed into aquatic environments, it is subject to new biogeochemical challenges, although
222 some of it remains as FDOM_H. In coastal waters, most organic matter is labile and RDOC mainly consists of
223 RDOC_t³⁰. Terrestrial FDOM_H may remain refractory until it is discharged into the ocean where it is subject to
224 further decomposition by the marine microbial community. However, in coastal regions, specifically when
225 high nutrient concentrations stimulated algal blooms (a condition known as eutrophication the mostly labile
226 DOC is produced which is then efficiently consumed by heterotrophic microorganisms. This locally produced
227 labile DOC has a priming effect^{91,92} on the river-discharged terrestrial RDOC, remobilizing the latter, at least

228 partly, for microbial utilization. Enhanced photochemical degradation of terrigenous DOM in the coastal ocean
229 also produces labile DOM. Thus, productive estuarine and coastal waters are often sources, rather than sinks
230 of atmospheric carbon dioxide^{93,94}. Therefore, reducing eutrophication has been proposed as an approach to
231 enhance carbon sequestration in coastal waters^{38,92}. This suggestion is supported by experimental results in
232 estuarine and offshore waters⁹⁵⁻⁹⁷.

233 234 **[H2] Influence of environmental conditions on microbial activity in the ocean water column**

235 Microbial transformation of DOM varies across various environmental gradients such as oxygen, temperature
236 and nutrients⁹⁸. The chemical composition of DOM also varies with ocean depth^{56,99} (FIG. 3). Water column
237 stratification leads to variations in temperature, dissolved oxygen, and pH, resulting in the accumulation of
238 RDOC in the water column¹⁰⁰. At elevated temperatures, heterotrophic respiration is enhanced, and the
239 efficiency of organic carbon utilization and transformation increases. Availability of dissolved oxygen also has
240 a significant impact on the preservation and degradation of DOM molecules^{101,102}. In oxygen minimum zones,
241 microbial degradation of DOM is reduced¹⁰³ and sulphur-containing compounds are formed, facilitating
242 RDOC production^{104,105}. Therefore, in these regions, carbon sequestration is relatively more dependent on the
243 fractionation of RDOC¹⁰⁶, compared to aerated water columns where POC export is dominant.

244 In the deep ocean, chemoautotrophic processes fix DIC, providing DOC sources other than those derived from
245 the decomposition of sinking particles¹⁰⁷. A recent study on the effect of hydrostatic pressure¹⁰⁸ indicates the
246 presence of a fraction of deep sea DOC that is not utilized due to the combined effect of high hydrostatic
247 pressure and low temperature. Low temperature and high hydrostatic pressure result in higher activation energy
248 requirements for microbial enzyme synthesis and reduced growth compared to depressurized conditions. The
249 addition and removal of deep sea RDOC is also driven by abiotic processes, such as hydrographic processes¹⁰⁹
250 and high temperature in hydrothermal vent systems¹¹⁰⁻¹¹².

251 252 **[H2] Beneath the ocean**

253 Aggregation processes in the ocean produce refractory particulate organic carbon (RPOC) in addition to
254 RDOC-coated POC, which is more resistant to decomposition and thus transported down to the sediment for
255 long-term storage³⁰. RPOC and RDOC in the sediment will undergo a series of chemical reactions under high
256 pressure and temperature (eg. pyrolysis), which over time could produce polycyclic aromatic hydrocarbons
257 (PAHs)^{113,114}. These compounds, in turn, would form a solid material of high molecular weight that is the
258 precursor to kerogen¹¹⁴ — the initial stage in the formation of fossil fuels. Once this hypothetical sequence of
259 processes is experimentally tested, we will be able to bridge the gap between marine organic matter from the
260 water column and kerogen in the Earth's crust.

261 262 263 **[H1] The microbial carbon pump and climate change**

264 Climate change will alter oceanic conditions which may influence the efficiency and capacity of the MCP and
265 other carbon pumps, to sequester carbon, potentially altering its role in climate regulation. Anthropogenic CO₂
266 emissions from fossil fuel use, deforestation, and cement production among others have increased atmospheric
267 CO₂ from 280 ppm in pre-industrial times to approximately 420 ppm in 2023 ([Global Monitoring Laboratory](#)).
268 There is consensus in the scientific community^{115,116} that emission reductions alone are insufficient and too
269 slow to avoid a global warming catastrophe. Thus, ONCE approaches or carbon dioxide removal technologies
270 are needed to remove significant (more than 10 gigatonnes per year (Gt/yr)) atmospheric CO₂ to limit global
271 warming to 2°C or to the best, 1.5°C above pre-industrial temperatures by the end of this century^{1,2}. Towards
272 this goal, the well-known mechanisms of the BCP, CCP, and SCP have all been addressed in the literature, but
273 the MCP is rarely discussed in this context. Therefore, we describe the role of the MCP in response to climate
274 change from a historical point of view with implications for current and future situations.

275 276 **[H2] The microbial carbon pump-driven refractory dissolved organic carbon pool regulates climate change**

277 The dynamics of the MCP-mediated RDOC reservoir can lead to either carbon sequestration or CO₂ release
278 over geological timescales, suggesting that the MCP is a two-way regulator of climate change (FIG. 4).
279 During interglacial periods of high atmospheric CO₂ concentrations, the oceans experienced increasing
280 warming, stratification, and deoxygenation¹¹⁷. In such cases, the MCP would become stronger¹¹⁸, leading to
281 an increase in RDOC production and accumulation, contributing to the mitigation of global warming. An
282 extreme example is the case in the Cryogenian period of the Neoproterozoic era (720 to 635 million years
283 before present), where a strong negative shift in the Carbon-isotopic composition of sedimentary carbonates,
284 together with a generally unchanged Carbon-isotopic composition of coexisting organic matter^{26,119,120},

285 indicates the presence of an unusually large RDOC reservoir at that time and therefore a very effective
286 MCP^{27,30,121,122}. The development of this large RDOC reservoir coincided with the extreme glacial event
287 extended to the equator known as “Snowball Earth”, which was widely confirmed in geological
288 records^{119,123,124}. In contrast, during glacial periods, the formation of sea ice leads to high salinity-driven
289 mixing that brings oxygen to depth and induces increased degradation of RDOC^{t125} with consequent CO₂
290 emission, which in turn mitigates global cooling.

291 During deglaciation, the CO₂ stored in the deep ocean was released into the atmosphere, leading to rapid
292 warming of the global climate^{126,127}. There was a rapid increase in atmospheric CO₂ concentrations,
293 accompanied by a significant decline in its Δ Carbon-14 (a measure of the age of a sample containing organic
294 materials by using the properties of the radioactive isotope of carbon that decays at a known rate, suggesting
295 that the CO₂ entering the atmosphere originated from old carbon which is depleted in Δ Carbon-14^{128,129}.
296 Considering the current situation of global warming and ocean stratification¹⁴, it is likely that the MCP is in a
297 strong and increasing phase in sequestering carbon in the ocean compared to the BCP.
298

299 **[H2] The microbial carbon pump and the amplification of the Milankovitch theory**

300 The Milankovitch theory is the foundation of climate evolution, which suggests that changes in the orbital
301 parameters of the Earth can cause variations in the amount and distribution of solar radiation reaching the
302 Earth's surface, which can then affect the Earth's climate over long periods. The eccentricity, that is, the
303 circularity of the Earth's orbit around the sun, theoretically contributes less than 0.1 % of the solar insolation
304 at the Earth's surface, yet the eccentricity cycle was the dominant cycle of climate change in the late
305 Quaternary¹³⁰. It is debatable how minor changes in orbital parameters could lead to incredibly large climate
306 changes in the glacial–interglacial cycle. Recent research on deep sea sediment records has shown that changes
307 in solar radiation can control the terrestrial nutrient input to the ocean, which, in turn, regulates the oceanic
308 carbon cycle. At long eccentricity maxima (~400 thousand years), summer insolation maximizes at low
309 latitudes, along with increased regional and global seasonality. This leads to an intensified global monsoon
310 and enhanced rainfall, increasing nutrient input to the ocean and surface productivity, which inevitably
311 increases carbon storage in the deep ocean and decreases marine inorganic carbon $\delta^{13}\text{C}$ (the ratio of ¹³C to ¹²C
312 in samples compared to the ratio in a standard)¹⁷. At long-eccentricity minima the opposite is true.

313 Such a hypothetical connection between eccentricity forcing and long-term changes in seawater carbon
314 isotopic signature has been simulated in a numerical model for the Miocene climate optimum during 14-17
315 million years. The modelling results showed that at eccentricity maxima or minima, the intensity of continental
316 weathering and nutrient supply lead to the minima and maxima of $\delta^{13}\text{C}$ in the ocean¹³¹. Therefore, the MCP
317 could amplify the effect of the Earth's orbital forcing on climate change by regulating the deep sea carbon
318 reservoir (RDOC-rich). The amplifying effect of the MCP has been manifested in climate events in the Earth's
319 early history, such as the severe glaciations in the Neoproterozoic Era, when the RDOC reservoir was
320 extremely large and could have played an even greater role in its climate than during the Quaternary¹²³. This
321 may explain why seawater $\delta^{13}\text{C}$ fluctuated with much larger amplitudes, reaching >10 per mille (‰) in the
322 Neoproterozoic and >1 ‰ in the early Cenozoic, while being only ~0.3-0.5 ‰ in the Pliocene-Pleistocene¹³²⁻
323 ¹³⁴. Therefore, variations in the size of the MCP-regulated oceanic carbon reservoir on geological time scales
324 may shed light on the eccentricity enigma of the Milankovitch climate theory.
325

326 **[H2] Modelling of the microbial carbon pump under climate changes**

327 Recent advances provide a unified hierarchical model that simultaneously predicts the preservation and
328 degradation of organic matter in geochemical systems¹³⁵. Additionally, there is a quantitative model for the
329 accumulation of DOM that explicitly resolves the ecological dynamics of consumers, which release multiple
330 pools of DOM. The model incorporates microbial uptake by a diverse community of consumers¹³⁶. A logical
331 and exciting next step would be an integration of the geochemical kinetics and ecological dynamics¹⁰¹.

332 Both ultra-high resolution mass spectrometry and high-throughput sequencing techniques provide semi-
333 quantitative information on the molecular composition of DOM and microbial species diversity. Advancing
334 our knowledge of the molecular structure of DOM and microbial functional diversity provides multi-layered
335 information. The complexity of the data enables machine learning algorithms and neural networks to recognize
336 patterns in RDOC, using the samples for training¹³⁷. The exponentially growing application of machine
337 learning and simulation models allows hypothesis testing. While experimental approaches have moved from
338 mainly in vivo to in situ, machine learning allows the testing of hypotheses in silico. The long-term stability
339 of RDOC was confirmed using a neural network model that mimics the encounter probability between
340 microorganisms and DOC in the ocean¹³⁸. This approach utilized assumptions about uptake and degradation
341 kinetics and simulated the age and concentrations of the oceanic DOC pool to match observations.

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[H1] Synergistic effects of the carbon pumps to maximize carbon sequestration

The first step towards maximizing ocean carbon sequestration would be a comprehensive understanding of the relevant processes involved in ocean carbon storage, including the interactions between the MCP, BCP, CCP, and SCP (FIG. 1 and 5). The BCP is key for ocean carbon sequestration. However, the majority of its organic carbon flux is ultimately respired to CO₂ and only a minor portion (less than 1%) reaches the seafloor for burial⁵. The production of calcium carbonate by the CCP is accompanied by the release of CO₂ into the water and atmosphere, creating a partial ‘counter’ effect to the uptake of CO₂. The SCP depends on the abiotic controls of the difference of CO₂ partial pressures between the atmosphere and surface seawater and downward transport processes. The four pumps occur simultaneously, with their relative importance depending on the environmental conditions, and thus they could potentially work synergistically for carbon sequestration. For example, the MCP may enhance the BCP through collision, binding and even coating of POC with RDOC molecules³⁰. The effect of the MCP on the CCP is twofold. In the water column, high molecular weight RDOC compounds or their aggregates could be the nuclei for carbonate precipitation^{24,139}. In the sediment, the MCP promotes precipitation of authigenic carbonate in situ within the sediment driven by microbially mediated processes that enhance alkalinity through bicarbonate and carbonate production via processes such as sulfate reduction, denitrification, and anaerobic oxidation of methane^{139,141,142}. In the surface waters, photosynthesis or application of alkaline minerals can enhance alkalinity and pH and thus uptake of CO₂ from the atmosphere¹⁴³. Fertilization with iron¹⁴⁴ and aluminum¹⁴⁵ promotes the efficiency of the BCP and CCP and thus the transformation of DOM within the MCP. Based on the above understanding, a concept of an integrated approach of the different carbon pumps — namely, BCP-CCP-MCP-SCP (BCMS) — is proposed¹⁹. This concept also encompasses the dual interpretation of BCMS as a ‘business continuity management system’ for effective risk reduction and control. The application of both BCMS interpretations (BCP-CCP-MCP-SCP and business continuity management system) would ensure that the implementation of ONCE is not only sensible and reasonable, but also compliant with legal standards.

[H2] Combined carbon pumps in sea-farming fields

Aquaculture areas are prone to eutrophication, hypoxia, and acidification in the bottom waters. To address these issues, artificial upwelling can be applied to bring nutrient-rich and high DIC waters from deeper levels to the surface. This promotes photosynthesis, which converts CO₂ into carbon biomass, increasing pH (> 8.5) and decreasing pCO₂ below atmospheric concentrations. This leads to CO₂ uptake (negative emission) into the ocean^{38,146}, thereby enhancing the SCP, BCP, MCP and CCP. The accompanying down-welling can bring oxygen-rich and high pH waters from the surface to the bottom, which mitigates hypoxia and favours CCP. For clarifying the role of CCP, we give an example in the upper Chesapeake Bay, where submerged aquatic vegetation reduces water pCO₂ to as low as 20 ppm and increases pH to as high as 10.2, leading to substantial calcium carbonate precipitation in areas with rich aquatic vegetation. In such cases, the CCP is a positive pump, as carbonate rather than bicarbonate is dominant, and even if there is CO₂ generated from calcium carbonate precipitation, it is not released to the atmosphere but serves as a substrate for further photosynthetic production¹⁴⁷. Artificial ocean alkalinity enhancements (OAE) can be applied in sea-farming areas to strengthen the efficiency of the combined four pumps.

[H2] Business continuity management system in wastewater treatment plants

The current wastewater treatment protocol has a serious disadvantage in terms of carbon sequestration, as it accelerates the release of CO₂ into the atmosphere by decomposing organic carbon into CO₂¹⁴⁸. The discharge of treated wastewater also acidifies the receiving coastal waters¹⁴⁸. Based on the BCMS principle, we propose to capture the DIC into calcium carbonate in treatment plants by using a strong base, such as sodium hydroxide from electrochemical production or minerals containing magnesium hydroxide and calcium hydroxide. The remaining residual carbon will be highly refractory in both solid (slurries) and dissolved (RDOC) forms, before being released into the environment. The wastewater effluent will also serve as an effective means for OAE via the spreading of river plumes and ocean mixing, and for mitigating ocean acidification in coastal waters¹⁴⁹. It remains to be seen whether the wastewater, which is mainly composed of RDOCt, will degrade after discharge. Additionally, the ecological effects of such an increase in ocean alkalinity need to be explored^{150,151}. Overall, the combined effects of BCMS are important for the design and implementation of ‘best practice’ carbon sequestration strategies (FIG. 6).

[H1] Conclusion

398 The MCP has become an active and evolving field of research, especially in its connection with RDOC
399 compounds and microbial diversity, and its role in the carbon cycle and climate change. In this Review, we
400 highlighted aspects where focused attention could significantly improve the understanding of MCP
401 mechanisms and its potential applications.

402 In the past decade, substantial advancements have enhanced our understanding of both microbial diversity and
403 DOM diversity. Such developments stem from metabolic and functional studies of key microbial groups, omics
404 and bulk analyses of DOM, and biogeochemical models.

405 With the accumulation of big data, more data science and systematic science approaches should be applied in
406 future studies of complex interactions between microorganisms and DOM. Multi-layer networks will help to
407 elucidate the functional relationships between microbial species and DOM molecules, to further illustrate the
408 formation of RDOC from microbial activities. These models, together with in silico simulations of long-term
409 climate change impacts, can showcase the mechanisms behind microbial RDOC production and the efficiency
410 of the MCP.

411 Faced with increasingly severe climate change issues, numerical simulations of past and present marine
412 microbial processes will not only help to understand the potential response of the MCP to climate change, but
413 also provide possible avenues for research on ocean carbon sequestration. Further insights into the microbial
414 and biogeochemical processes associated with the MCP, their interactions with the other carbon pumps, and
415 their potential applications for carbon sequestration would be maximized through the production of a virtual
416 ‘digital twin’ replicating real-world scenarios. Additionally, machine learning and numerical models can
417 generate geological event-based scenarios. The outcome of these approaches would also help the evaluation
418 of ONCE or marine CDR techniques and future projections of ocean carbon storage.

419 Microbially-driven RDOC has acted as a buffer against climate change throughout Earth’s history. Studies
420 over the past decade have provided a wealth of data on both microbial biomolecular and ecological processes,
421 as well as on RDOC compounds and their recalcitrance. However, critical gaps persist in understanding the
422 link between different mechanisms, their interactions, and their biotic and abiotic controls. To address these
423 gaps, real-world observations, experimental and modelling studies, and digital twin simulations are needed. It
424 is particularly desirable to understand the processes and their environmental boundary conditions to promote
425 the synergistic effects of the four ocean carbon pumps (BCP-CCP-MCP-SCP) for efficient carbon
426 sequestration. Linking microbial taxonomic and functional diversity, not only with the chemical diversity of
427 DOM but also with the processes and synergistic interactions within the BCMS offers a systematic approach
428 to understand microbial-driven carbon cycling and carbon sequestration.
429

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778 Figures

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782 Fig. 1. Marine carbon cycling and the major processes and mechanisms involved

783 DOC-dissolved organic carbon, LDOC-labile DOC, RDOC-refractory DOC, POC-particulate organic carbon,

784 CCN-Carbonate Condensation Nuclei, CCD: Carbonate Compensation Depth, AE-alkalinity enhancement.

785 The solubility carbon pump (SCP): Driven by the difference of CO₂ partial pressures between the atmosphere and
786 surface waters, exchanges of CO₂ occurs through dissolution into water or release into the air. Generally, the SCP
787 refers to the pumping of CO₂ from the atmosphere into the ocean driven by abiotic processes such as lowering
788 temperature and downward mixing. Biological carbon pump (BCP): The BCP refers to a series of biogeochemical
789 processes that transport organic carbon (mainly POC) from the surface to the ocean interior. The BCP flux depends
790 on sinking processes and attenuates sharply with depth, only a tiny portion of the carbon fixed in the euphotic zone
791 reaches the bottom of the sea for long term carbon sequestration. Microbial Carbon Pump (MCP): The MCP refers
792 to a series of microbial physiological and ecological processes which transform LDOC to RDOC that can stay in
793 the water column for a long term (up to thousands of years), serving as a carbon sequestration mechanism. MCP
794 occurs throughout the ocean water column. Carbonate counter pump (CCP): The CCP refers to the carbonate's
795 formation and deposition processes. Because of the accompanied CO₂ release during the carbonate's formation
796 process, it is called a "counter" pump. However, the release of CO₂ in the real world is not really at equivalent level
797 as indicated in the chemical equation (see text). On the continental shelf, CCP can constitute long term carbon
798 sequestration if the precipitated carbonates are buried in the sediment. In oceanic waters, carbonates precipitate at
799 the surface water (mainly by microalgae such as *Coccolithopherids*) but dissolves at the CCD layer. In the sediment,
800 authigenic carbonate occurs when alkalinity enhanced by microbial processes (see text). The RDOC coating effect
801 refers to the processes that RDOC molecules adhere to POC particles, preventing somewhat further degradation of
802 POC.

803

804

805 **Fig. 2. Microbial-dissolved organic matter complex networks.** Microbial-dissolved organic matter (DOM) complex
806 networks contain two types of nodes: the microbial nodes and the DOM nodes. Connections are made between nodes
807 based on correlations of datasets. Microbial diversity can be analysed with 16S rRNA amplicons, metagenomics,
808 metatranscriptomics, and metaproteomics. As for the DOM components, ultra-high resolution mass spectrometry can be
809 used to determine compositional diversity, and nuclear magnetic resonance spectroscopy can be used to reveal structural
810 diversity. These individual connections collectively form the overall network pattern. Analysis based on key nodes unveils
811 the relationships between refractory dissolved organic carbon (RDOC) molecules (for example, carboxyl-rich alicyclic
812 molecules (CRAMs)) and key microbial populations (for example, Kyoto encyclopaedia of genes and genomes (KEGG)
813 modules, including metabolic pathways, signalling pathways, complexes, functional sets, etc.), thus revealing the
814 connections between them. Figure adapted with permission from refs^{25,74}. On the right panel, red lines represent positive
815 associations and blue lines represent negative associations between RDOC molecules and microbial populations.

816

817

818 **Fig. 3. The microbial carbon pump in changing environments.** Microbial carbon pump (MCP) processes (orange)
819 occur in environments from soil to seawater, and from coastal to oceanic waters. Changes in environmental conditions
820 such as nutrients and dissolved oxygen concentrations have impacts on the production and transformation of refractory
821 dissolved organic carbon (RDOC) (red arrows). For example, eutrophication could ultimately lead to a reduction in the
822 efficiency of the MCP, because heterotrophic microbes are fuelled by indigenous labile organic carbon, together with
823 high nutrients, they become not only larger in population but also more energetic and capable of respiration, and some
824 terrestrial RDOC compounds become no more refractory in front of them. Such priming effects lead to high rates of
825 biodecomposition and following photochemical degradation.

826 The BCP, depicted by the green funnel-shaped icon, involving CO₂ absorption by phytoplankton via photosynthesis and
827 subsequent organic matter generation within the food web. This organic matter partly feeds into the MCP
828 RDOC in oceanic surface seawater is subject to photodegradation. Chemoautotrophic processes that occur in the deep
829 ocean Chemoautotrophic processes, such as the conversion of ammonia to nitrate by nitrifying bacteria in the deep sea
830 and the oxidation of hydrogen sulfide at hydrothermal vents by sulfur-oxidizing bacteria are crucial for energy conversion
831 in light-independent ecosystems and elemental biogeochemical cycling. Microorganisms involved in these
832 chemoautotrophic processes also could contribute to the RDOC pool. RDOC can aggregate, contributing to particulate
833 organic carbon burial in the sediment, and could be associated with the formation of kerogen which is the source of oil.
834
835

836 **Fig. 4. The refractory dissolved organic carbon pool driven by the microbial carbon pump is a two-way**
837 **regulator of climate change.** During a warm period (interglacial), the deep-sea refractory organic carbon
838 (RDOC) pool gradually builds up below the thermocline acting as a cooling mechanism for the atmosphere (left). On the
839 other hand, during a glaciation period, mixing of the water column remobilizes the oxidation and outgassing of
840 intrinsically recalcitrant RDOC acting as a warming mechanism for the atmosphere (right). Both processes mitigate
841 climate fluctuations. The trend of atmospheric carbon dioxide variation in the past 800,000 years has an obvious glaciation
842 cycle process of 100,000 years. The geological history of carbon dioxide is known from the Antarctic ice core record.
843 (Data source: [Paleo data search](#)). The dashed red line in the carbon dioxide (CO₂) record represents a hypothetical scenario
844 where CO₂ release during deglaciation is delayed in the absence of the RDOC pool; values are calculated with a 0.95-
845 fold multiplier of the original value during deglaciation to indicate a hypothetical scenario^{127,152,153}. ppm, parts per million;
846 Kyr, thousand years.
847
848

849 **Fig. 5. Differences between the biological carbon pump, the microbial carbon pump and the carbonate counter**
850 **pump.** The biological carbon pump (BCP) carries a fraction of the photosynthetically fixed carbon to the deep ocean to
851 be buried in the sediment, where it is removed permanently from the ocean. Because of this the BCP can be termed a
852 'one-way pump'. The microbial carbon pump (MCP) can be termed as a 'two-way pump' as it stores the refractory
853 dissolved organic carbon (RDOC) within the ocean for a long time before it is remineralized to carbon dioxide (CO₂)
854 and potentially available for outgassing to the atmosphere. The carbonate counter pump (CCP) is a 'counter pump'
855 because a substantial fraction of CO₂ may be released to the atmosphere (depending on the pH of the water and the
856 partial pressure of CO₂ of the atmosphere) when a mole of carbonate is precipitated.
857
858

859 **Fig. 6. An integrated approach for maximum carbon sequestration in the ocean.** A proposed integrated approach of
860 the different carbon pumps — the biological carbon pump (BCP), the carbonate counter pump (CCP), the microbial
861 carbon pump (MCP) and the solubility carbon pump (SCP) — named BCMS approach, is proposed to maximize the
862 sequestration of carbon in the ocean. Different strategies include estuarine remediation⁹², whereby reduction of terrestrial
863 nutrient inputs can enhance the overall efficiency of the four pumps by reducing respiration and water acidification,
864 enhancing both refractory dissolved organic carbon (RDOC) and net carbon burial through biotic and abiotic processes;
865 artificial upwelling in sea farming fields^{154,155}, aimed at bolstering the production and burial output of the BCP, and
866 augmenting the MCP and sediment CCP output, while improving bottom water quality; wastewater treatment plant
867 (WWTP)-based ocean alkalinity enhancement (OAE)¹⁴⁹ can drive the SCP in the surface ocean, and increase alkalinity
868 in nearshore areas, thereby promoting the SCP and further carbonate deposition by the CCP, while the DOC in sewage is
869 partially converted to RDOC via the MCP¹⁵⁶. Lastly, the application of iron-aluminium fertilization can mitigate iron
870 limitation of phytoplankton growth and promote particulate organic carbon (POC) sinking, thus enhancing the efficiency
871 of the BCP in oceanic waters. authigenic carbonate are formed by microbial activity or chemical changes in sediments,
872 as described in the " Synergistic effects of the carbon pumps to maximize carbon sequestration " section.
873
874