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Dissecting the structural and chemical determinants of the 'open-to-closed' motion in the mannosyltransferase PimA from mycobacteria.

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ABSTRACT

The phosphatidyl-mvo-inositol mannosyltransferase A (PimA) is an essential peripheral membrane glycosyltransferase that initiates the biosynthetic pathway of phosphatidyl-myoinositol mannosides (PIMs), key structural elements and virulence factors of Mycobacterium tuberculosis. PimA undergoes functionally important conformational changes, including (i) α -helix-to- β -strand and β -strand-to- α -helix transitions, and (ii) an 'open-to-closed' motion between the two Rossmann-fold domains, a conformational change necessary to generate a catalytically competent active site. In previous work, we established that GDP-Man and GDP stabilize the enzyme and facilitate the switch to a more compact active state. To determine the structural contribution of the mannose ring in such activation mechanism we analyzed a series of chemical derivatives, including mannose-phosphate (Man-P) and mannose-pyrophosphateribose (Man-PP-RIB), and additional GDP derivatives, as pyrophosphate-ribose (PP-RIB) and GMP, by the combined used of X-ray crystallography, limited proteolysis, circular dichroism, isothermal titration calorimetry and Small Angle X-ray Scattering methods. Although the βphosphate is present, we found that the mannose ring, neither covalently attached to phosphate (Man-P) nor to PP-RIB (Man-PP-RIB), does promote the switch to the active compact form of the enzyme. Therefore, the nucleotide moiety of GDP-Man, and not the sugar ring, facilitates the 'open-to-closed' motion, with the β -phosphate group providing the high affinity binding to PimA. Altogether, the experimental data, contribute to a better understanding of the structural determinants involved in the 'open-to-closed' motion observed not only in PimA, but also visualized/predicted in other glycosyltransfeases. In addition, the experimental data might prove useful for the discovery/development of PimA and/or glycosyltransferase inhibitors.

INTRODUCTION

It is now generally accepted that proteins do not just occupy one single state and biological functions, i.e. substrate recognition, binding, stabilization of transition states and product release, are ultimately rooted in the coordinated physical motions of the atoms, localized protein regions, motifs, and/or protein domains in a wide range of timescales. Protein dynamics can be defined as any time-dependant change in atomic coordinates including both equilibrium fluctuations and non-equilibrium changes (1). A hypothesis proposed over 30 years ago by Frauenfelder and colleagues and more recent experimental data suggest that protein undergoes conformational changes leading to an ensemble of conformations sampled over the time in the multidimensional energy landscape (1, 2, 3, 4). The landscape defines the amplitude and directionality, and the timescale of the motions to adopt substates (1). Therefore, understanding the relationship between structure, catalysis and function of proteins often requires the consideration of their dynamic behavior/nature, understood as the ensemble of conformations sampled in a multidimentional/intricate energy landscape. These intrinsic conformational transitions can be classified by the amplitude of motions. Submicrosecond fluctuations arising from side chains rotations and small loops movements are commonly observed during substrate recognition and catalysis. In addition, enzyme function can involve slower and larger-amplitude collective motions, as fold-switches or domain motions, between a relatively small number of substates (5). Correlations between the conformational fluctuation timescale and the catalytic turnover have been empirically stablished for a variety of enzymes – e.g. CypA (3) and HIV-1 protease (6). Interestingly, the rate of conformational change has been shown to coincide with the rate limiting step, such as product release (3).

The phosphatidyl-*myo*-inositol mannosyltransferase A (PimA), is an essential membrane glycosyltransferase (GT) that initiates the biosynthesis of phosphatidyl-*myo*-inositol mannosides (PIMs), lipomannan (LM) and lipoarabinomannan (LAM), key structural

elements and virulence factors of Mycobacterium tuberculosis (7, 8, 9). PimA catalyzes the transfer of a Manp residue from GDP-Man to the 2-position of phosphatidyl-myo-inositol (PI) to form phosphatidyl-myo-inositol monomannoside (PIM₁) on the cytoplasmic side of the plasma membrane (see the chemical structure of Ac₂PIM₆ and the scheme of the reaction catalyzed by PimA in Figure S1; 7, 10). PimA is an amphitropic GT that belongs to the GT-B superfamily and is composed of two Rossmann-fold domains separated by a central deep cleft that includes the catalytic center (11, 12, 13, 14). The unliganded crystal form of PimA presents four monomers in the asymmetric unit, capturing two different conformational states of the enzyme. The two states are structurally similar to each other, displaying an 'open' conformation. However, an α -helical hairpin consisting of helices $\alpha 4$ (residues 134–145) and α 5 (residues 149–157) folds back against the N-terminal domain in one conformational state (the 'compact inactive' state, thereafter), whereas in the other state it displays an extended conformation (the 'extended inactive' state, thereafter) that is partially disordered and protrudes away from the N-terminal domain core (Figure 1; 16, 17). The structural comparison of the unliganded form with that of PimA in complex with GDP-Man or GDP revealed important conformational changes. Specifically, (i) the occurrence of fold-switch β strand–to– α -helix and α -helix–to– β -strand transitions in the N-terminal domain of the enzyme (residues 118–163; 15, 17), and (ii) an domain rearrangement from an 'open' ('active extended' state, thereafter) to a 'closed' ('active compact' state, thereafter) state (11, 18, 19). Moreover, the fold-switching is a functional event that activates PimA, since a PimA mutant locked in both extended and compact unliganded conformations is inactive, while a mutant locked in the substrate-bound conformations is active (15).



Figure 1. Current structural snapshots of PimA along the catalytic cycle. The crystal structure of the unliganded form of PimA (PDB ID 4NC9) trapped two different conformations of the enzyme, the 'inactive compact' and 'inactive extended'. An α -hairpin moves away from the core of the N-terminal domain. The crystal structure of PimA in complex with GDP-Man (PDB ID 2GEJ) reflects the 'active compact' state of the enzyme. The transition to the active state involves a fold-switch event localized in a region of the N-terminal domain. It is worth noting that both inactive forms of PimA crystallized in an 'open' conformation with respect of the N- and C-terminal domains, whereas the active conformation crystallized in a 'closed conformation'.

The large amount of structural evidences places us in an unprecedented position to study the donor substrate-mediated mechanism of activation of PimA. Here we decipher the structural contribution of GDP-Man to the activation mechanism of PimA, by the combined use of nucleotide sugar chemistry, biochemistry, X-ray crystallography, small-angle X-ray scattering, circular dichroism and isothermal titration calorimetry.

EXPERIMENTAL PROCEDURES

Materials and Methods

Guanosine 5'-diphosphate-D-mannose (GDP-Man), guanosine 5'-diphosphate (GDP), guanosine 5'-monophosphate (GMP), guanosine (GNO), guanine (GUA), mannose 1phosphate and elastase protease were purchased from Sigma. Recombinant PimA from M. smegmatis (MSMEG 2935) was produced in *Escherichia coli* and purified to apparent homogeneity as previously described (15, 18). The reaction flasks and other glass equipment were heated in an oven at 130 °C overnight and assembled in a stream of argon. All reactions were monitored by TLC on silica gel 60 F254; the positions of the spots were detected by λ =254 nm UV light or by spraying with either a 5 % solution of phosphomolybdic acid in ethanol or Mostain solution. Column chromatography was carried out in a Buchi 800 MPLC system or a Combiflash apparatus by using silica gel 60 microns and with solvents that were distilled prior to use. Melting points were uncorrected. Purification by semipreparative HPLC (column Atlantis[®] DC18 5 µm, 19×100 mm, flow: 12.5 mL min⁻¹) was carried out in a Waters 515 pump system with photodiode array (PDA) detection. ¹H and ¹³C NMR spectra were recorded on Bruker Avance 400 MHz or AVANCE II 300 MHz instruments in the stated solvent. Chemical shifts are reported in ppm (δ) relative to CHCl₃ (δ =7.26) in CDCl₃. NMR assignments were made by using standard 2D experiments. Optical rotations were recorded on a JASCO DIP-370 polarimeter. Elemental analyses were performed on a PerkinElmer 240B microanalyzer or with a PerkinElmer 2400 instrument.

MsPimA•GMP crystallization and Data Collection

The best crystals of the *Ms*PimA•GMP form were obtained by mixing 2.0 μ l of GSGA-PimA (10 mg ml⁻¹) in the presence of 1 mM GMP, with 2.0 μ l of a mother liquor of 18% PEG 8000, 200 mM calcium acetate and 50 mM HEPES pH 7.5. Crystals appeared after 2-3 days and

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grew as rods. Prior to data collection, crystals were transferred to a cryoprotectant solution (25% glycerol in the mother liquor) and immediately cryo-cooled in liquid nitrogen. Crystals of *Ms*PimA•GMP belong to the orthorhombic space group $P2_12_12_1$ and have 1 molecule per asymmetric unit, corresponding to a Matthews coefficient of 2.31 Å³ and a solvent content of 46.7%. X-ray diffraction data from a single crystal were collected on Proxima 1 beamline (λ = 0.98011 Å) at Soleil (France) equipped with a CCD ADSC Quantum 315r system detector, and processed with program XDS (20). A total of 300 diffraction images were collected with a 0.5° rotation between images.

MsPimA•GMP structure determination and refinement

The crystal structure of the *Ms*PimA•GMP complex was solved by molecular replacement method with the program Phaser and PHENIX suite (21, 22), using the atomic coordinates of *Ms*PimA•GDP-Man complex as the search model (PDB ID 2GEJ: 11). Model rebuilding was carried out with the CCP4 suite (23). The final manual building was carried out with Coot (24) and refined with phenix.refine (25). The structure was validated by MolProbity (26). Data collection and refinement statistics are shown in Table S1. The atomic coordinates and structure factors have been deposited in the Protein Data Bank, (PDB ID 5BRS).

Chemical synthesis of Man-PP-RIB and PP-RIB

Please see Supporting Information for details.

Limited Proteolysis of MsPimA

Recombinant purified *Ms*PimA (25 μ g) was incubated with 0.10 μ g of elastase in 100 μ l of 50 mM Tris-HCl pH 7.5, in the presence of substrates for 0–90 min at 37 °C. Aliquots of 12 μ l were mixed with 3 μ l of 250 mM Tris-HCl pH 6.8, 10% SDS, 50% glycerol, 500 mM DTT

and 0.01% bromophenol blue at the indicated times. Samples were boiled for 3 min and run onto a NuPAGE[®] 4-12% gel (Invitrogen). Protein bands were visualized by staining the gel with SimplyBlueTM SafeStain (Invitrogen).

Near-UV Circular Dichroism analysis

Spectra were adquired in a J-810 CD spectropolarimeter (Jasco Corp., Tokio, Japan) by using Hellma 105.200-QS quartz cuvettes with a 1 cm optical path. Spectra were recorded in a continuous mode with 1 nm band width, one second response and a scan speed of 100 nm/min⁻¹. Samples were 10 μ M *Ms*PimA in 10 mM Tris-HCl pH 7.5. Substrates were added in a 1:10 ratio and 25 scans were accumulated to obtain the final spectra, which were further corrected for the baseline signal. Spectra were recorded in the 250-300 nm range at 20°C.

Temperature scans

Spectra were adquired in a J-810 CD spectropolarimeter (Jasco Corp., Tokio, Japan) by using Hellma 110-QS quartz cuvettes with a 1 mm optical path, by using a Peltier thermal device, allowing the temperature control during the experiments. Spectra were recorded in a continuous mode with 1 nm band width, one second response and a scan speed of 100 nm/min⁻¹. Samples were 4.61 μ M *Ms*PimA in 10 mM Tris-HCl pH 7.5. Substrates were added in a 1:10 ratio. In this case, thermal dependancies of the ellipticity were monitorized in the range from 10°C to 90°C at 222 nm. Temperature was increased stepwise by 1°/min. (Table 1).

Small Angle X-ray Scattering Measurements

Synchrotron X-ray diffraction data for purified recombinant *Ms*PimA in its unliganded form and in the presence of GDP-Man, GDP, GMP, PP-RIB, Man-P and Man-PP-RIB were

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collected in batch mode. The sample volume loaded was 30 µL (1.5 mm diameter capillary with 10 µm wall thickness). Data were collected on a pixel Pilatus 2M detector at Diamond Light Source B21 beamline (UK). The scattering patterns were measured as an accumulation of frames taken as 18x0.5-sec exposures for protein samples at a minimum of three different protein concentrations ranging from 4 to 8 mg/ml. To check for radiation damage, the 18 frames were compared as a time-series; no radiation damage was observed. Using the sample-to-detector distance of 3914 mm, the range of momentum transfer values is $0.004 < q < 0.37 \text{Å}^{-1} (q = 4\pi \sin(\theta)/\lambda$ where 2 θ is the scattering angle and $\lambda = 1$ Å is the X-ray wavelength). Data were processed using standard procedures by the program packages ScÅtter (developed by Rob Rambo at Diamond Light Source) and PRIMUS (27). The forward scattering (I(0)) was evaluated using the Guinier approximation (28) assuming the intensity is represented as $I(q) = I(0)\exp(-(qR_g)^2/3)$ for a very small range of momentum transfer values ($q < 1.3/R_g$). The maximum dimensions (D_{max}), the interatomic distance distribution functions (P(r)), and the radii of gyration (R_g) were computed using GNOM (29) (see Table S2 for details).

Isothermal titration calorimetry

Ligand binding to *Ms*PimA was assayed using the high precision VP-ITC system (MicroCal Inc.) as previously described (11) with the following modifications. The ITC cell (1.4 ml) contained 10 μ M *Ms*PimA in 25 mM Tris-HCl pH 7.5 and 150 mM NaCl, and the syringe (150 μ l) contained 150 μ M of GDP, 500 μ M of GMP, Man-P, Man-PP-Rib or PP-Rib, in the same buffer. Sample solutions were thoroughly degassed under vacuum and each titration was performed at 25°C by one injection of 2 μ l followed by 29 injections of 10 μ l with 210 seconds between injections using a 290 rpm rotating syringe. Raw heat signal collected with a 16 s filter was corrected for the dilution heat of the ligand in the PimA buffer and normalized

to the concentration of ligand injected. Data were fit to a bi-molecular model using the OriginTM software provided by the manufacturer.

RESULTS

The crystal structure of MsPimA in complex with guanosine 5'-monophosphate

We were able to solve the crystal structure of PimA from Mycobacterium smegmatis (MsPimA; UniProt ID A0QWG6) in complex with guanosine 5'-monophosphate (GMP) at 2.6 Å resolution (Data collection and refinement statistics are shown in Table S1; Figure 2A). MsPimA crystallized in an active compact conformation, similar to that observed in the MsPimA•GDP (PDB ID 2GEK) and MsPimA•GDP-Man (PDB ID 2GEJ) complexes (r.m.s.d. of 0.4 Å for 361 equivalent residues). In contrast, the structural comparison of the MsPimA•GDP complex with the inactive compact and extended unliganded forms of MsPimA revealed the local reshuffling of secondary structure elements within the flexible segment (residues 118–163) in the N-terminal domain (Figure 1; r.m.s.d. of 5.6 Å and 9.2 Å for 366 and 358 equivalent residues, respectively; 15). The GMP molecule is clearly visible in the electron density map, located at the interface between the N- and C-terminal domains (Figure 2B and 2C; see the Final (2Fo-Fc) electron density map for MsPimA-GMP complex (contoured at 1σ) in Figure S2). The guanidyl heterocycle binds to a hydrophobic pocket defined by the connecting loop $\beta 1-\alpha 1$ from the N-terminal domain (residues Val13—Gly16), the ends of strands $\beta 8$ and $\beta 9$, and the connecting loop $\beta 10 - \alpha 9$ from the C-terminal domain. The ribose oxygens O2' and O3' interact with the carboxylate group of Glu282, a conserved glutamate residue of the EX₇E motif, present in some members of the GT-B-fold superfamily (12, 30). Interestingly, the α -PO₄ of GMP displays a different conformation when compared to the position observed in the PimA•GDP and PimA•GDP-Man crystal structures (Figure 2C; 18). The α -PO₄ O1' makes new hydrogen bond with the main chain of Gly16, almost reaching the position of the β -PO₄ O3' in GDP and GDP-Man (Figure 2C). The catalytic loop (residues 268 to 277) adopts a different structural arrangement closer to the α -PO₄ (r.m.s.d. of 0.9 Å and 1.0 Å for 10 equivalent residues in PimA•GDP and PimA•GDP-Man, respectively).





Figure 2. The crystal structure of PimA in complex with GMP. *A*. Overall structure of *Ms*PimA in complex with GMP. The N-terminal and C-terminal domains are shown in yellow and orange respectively. *B*. Structural superposition between *Ms*PimA•GMP and *Ms*PimA•GDP-Man (grey) complexes. *C*. Structural superposition of GDP-Man and GMP as observed in the crystal structures of the corresponding crystal structures.

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The main-chain oxygen of Ser275 makes a hydrogen bond with the structural water molecule W1, which in turn interacts with the α -PO₄ O3', and the side-chain of Glu274 (11). The ε -amino group of the essential Lys202 lateral chain, which in the GDP and GDP-Man complexes established an important polar interaction with the β -PO₄ O2', makes now a hydrogen bond with the α -PO₄ O2'. In contrast, the lateral chain of another essential residue Arg196, also involved in the interaction with the β -PO₄, moves away from the active site (Figure 2C). Thus, the conformational changes observed in the active site of the *Ms*PimA•GMP complex raise the question about the precise structural elements of the nucleotide sugar donor involved in the conformational transitions observed in PimA.

Chemical synthesis of GDP-Man derivatives

We dissected the GDP-Man structure generating truncated versions as an approach to determine, what are the structural component/s within the native donor substrate structure responsible for the (i) fold-switch from inactive extended to active extended, (ii) motion from extended active to compact active and (iii) stabilization of the active compact conformation. Thus, we selected a series of GDP-Man derivatives: those containing the nucleobase ring, including GDP, GMP, guanosine (GNO) and guanine (GUA); and those containing the mannose ring, mannose-phosphate (Man-P) and mannose-pyrophosphate-ribose (Man-PP-RIB). In addition, we have synthetized pyrophosphate-ribose (PP-RIB), which connects to the mannose residue through the β -phosphate and to the nucleobase through the ribose ring (Figure 3A).

The synthetic derivatives Man-PP-RIB and PP-RIB were designed as β -*O*-methyl glycosides at the ribose moiety to ensure a well-defined β -anomeric configuration, mimicking the electronic and conformational properties of the natural substrate and minimizing any undesired side-interactions and steric hindrance. Two different synthetic approaches were

> used for the preparation of the syntetic derivatives. In the case of Man-PP-RIB the key reaction involved the phosphate-phosphate coupling between the activated forms of Man-P and P-RIB (see the chemical synthesis of Man-PP-RIB, in Scheme S1 of the Supplementary Information). Commercially available D-ribose was converted into protected compound 1 that, after removal of the trityl group, furnished 2 in good yield. The free hydroxyl at C-5 was phosphorylated using dibenzyl N.N-diisopropylphosphoramidite and the resulting phosphite ester was oxidized in situ with 'BuOOH giving compound 3 that was deprotected and activated as imidazolyl phosphate yielding 4. Finally, phosphate coupling of activated compound 4 with α -D-mannose-1-phosphate using MgCl₂ afforded Man-PP-RIB in good overall yield. PP-RIB was prepared through a sequential phosphorylation reaction using the phosphoramidite/oxidation method (see the Chemical synthesis of PP-RIB in Scheme S2 of the Supplementary Information). Commercially available D-ribose was converted into protected compound 5 and subsequently, the free hydroxyle at C-5 was phosphorylated using the phosphoramidite/^tBuOOH technique furnishing compound 6 which was debenzylated Pd-catalyzed hydrogenation to give 7. Treatment of 7 with N,Nthough diisopropylphosphoramidite / 'BuOOH to give 8, followed by removal of the isoproplidene group and debenzylation through hydrogenolisis furnished PP-RIB in good overall yield.

The role of the β -PO₄ in the generation of the active compact conformation of MsPimA

First, we determined the impact of the nucleobase derivatives interaction on the conformation of *Ms*PimA by limited proteolysis. We have previously demonstrated that in the presence of GDP, which comprises the nucleobase, the ribose moiety and the α -PO₄ and β -PO₄, *Ms*PimA is barely cleaved, even after 90 min incubation with elastase (18).

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Figure 3. Endoproteolytic cleavage of *Ms***PimA in the presence of GDP-Man derivatives.** *A.* - Chemical structure of GDP-Man derivatives. *B.* SDS-PAGE showing the elastase cleavage profile for *Ms*PimA preincubated with GDP-Man derivatives.

Here we see, as depicted in Figure 3B, that when MsPimA was incubated with GMP, in which only the α -PO₄ of GDP is present, the enzyme was mostly protected from the degradation by the protease. GNO, lacking the pyrophosphate group of GDP, only partially protected MsPimA from the action of the elastase. In contrast, MsPimA was rapidly degraded after incubation with the nucleobase GUA, and PP-RIB showing a similar proteolytic profile to that observed with the unliganded form of the enzyme (Figure 3B). It is worth noting that we have previously microsequenced the two predominant small species of 23 and 15 kDa revealing the sequences (i) SAMRS, located in $\alpha 9$ and (ii) SFADA, in the connecting loop β 7- β 8 at the junction between N- and C-terminal domains. Interestingly, the α 9 helix contains two critical residues involved in donor substrate recognition: Asp253, which interacts with N2 of the guanidyl group of GDP-Man, conferring MsPimA its specificity for the nucleoside; and Lys256, which participates in ribose binding (11, 18; Figure 3). Secondly, we studied by near-UV circular dichroism (CD) the tertiary structural changes of PimA induced by the nucleotide derivatives. The near-UV CD spectrum of MsPimA in its unliganded form in solution exhibited a broad positive signal characteristic of aromatic side chains, namely phenylalanines, tyrosines and tryptophans, localized in the protein (Figure 2B). The addition of GDP induced a gain of the tertiary structure constrains on aromatic residues, as indicated by an increment of the dichroic signal (Figure 4A). The addition of GMP also increased the dichroic signal on MsPimA, however not as high as that observed for the PimA•GDP complex. The incubation of MsPimA with GNO generated small, but still detectable differences in the near-UV CD profile. In contrast, the addition of the nucleobase GUA or that of PP-RIB, did not change the spectrum.



Figure 4. Near CD of *Ms***PimA in the presence of GDP-Man derivatives.** *A*. Near-UV CD spectra for the unliganded form of *Ms*PimA (black) and the *Ms*PimA•GDP (red), *Ms*PimA•GMP (light blue), *Ms*PimA•GNO (pink), *Ms*PimA•PP-Rib (purple) and *Ms*PimA•GUA (yellow) complexes. *B*. Near-UV CD spectra for the unliganded form of *Ms*PimA (black) and the *Ms*PimA•GDP-Man (blue), *Ms*PimA•Man-PP-Rib (orange) and *Ms*PimA•Man-P (green) complexes.

Thermal unfolding followed by CD at a wavelength of 222 nm showed similar apparent melting temperatures ($T_{\rm M}$) of 37.0, 36.6, 36.4, 36.4 and 36.2 °C, for the *Ms*PimA•GMP, *Ms*PimA•GNO, *Ms*PimA•GUA, *Ms*PimA•PP-RIB and the unliganded form of *Ms*PimA, respectively (Table 1). An important observation was that the $T_{\rm M}$ of *Ms*PimA•GDP complex was 46.4 °C, indicating that this state is *ca.* 10 °C more stable than *Ms*PimA or the *Ms*PimA•GMP, *Ms*PimA•GNO, *Ms*PimA•GUA complexes. PimA alone displayed a van't Hoff enthalpy change (Δ H) of about 70 kcal/mol. The addition of GMP, GNO, GUA or PP-RIB had no effect on *Ms*PimA Δ H parameters. The addition of GDP significantly increased the Δ H to 155 kcal/mol. The *K_d* determination by ITC revealed that GDP-Man and GDP bind to *Ms*PimA in enthalpy-driven reactions with dissociation constants, *K_d* of 0.23 µM and 0.03 µM, respectively (11). In contrast, no binding of GMP, GNO, GUA could be detected to *Ms*PimA (*K_d* > 100 µM) under the same experimental conditions (see Experimental Procedures section). Therefore, there is a clear/direct correlation between the chemical complexity of the derivative and the gain of ternary structure consistent with their binding

interactions in MsPimA. Only the GDP derivative binding, which contains the β -PO₄,

triggered the stabilization of MsPimA (Table 1).

Table 1. Unfolding parameters of free and *Ms*PimA bound to GDP-Man derivatives, measured by CD.

	Tm (ºC)	∆H (kcal/mol)
Unliganded	36.3	67.3
GDP-Man	45.5	165.7
GDP	47.5	155.6
GMP	37.0	70.1
GNO	36.6	63.4
GUA	36.4	66.9
Man-P	36.5	59.0
Man-PP-Rib	36.2	66.3
PP-Rib	36.4	64.5

The mannose residue is dispensable for the generation of the active compact conformation of MsPimA

How does the sugar ring binding impact the conformation of *Ms*PimA? The crystal structure of the *Ms*PimA•GDP-Man complex revealed that the mannosyl moiety of GDP-Man is stabilized within the active site cleft through several hydrogen bonding contacts mainly located in the catalytic loop: O3 makes a hydrogen bond with the main-chain of Ser275; the O4 makes important hydrogen bonding interactions with the lateral chain of Glu274 and the main chain of Phe276; whereas the O6 engages the nitrogen atom of the main chain of Ile278 (11). As depicted in Figure 3B, while *Ms*PimA was completely protected from proteolytic digestion by GDP-Man, as also observed for GDP, the enzyme was not protected from elastase degradation after incubation with Man-P or Man-PP-RIB, where the β -PO₄ is present. The near-UV CD spectra of the *Ms*PimA•GDP-Man complex revealed clear differences in the dichroic signal with respect to the unliganded form of *Ms*PimA reflecting variations in the ternary structure of the enzyme. Interestingly, the addition of Man-P or Man-PP-RIB, did not change the spectrum when compared to that of the unliganded form of *Ms*PimA (Figure 4B). The thermal unfolding profile of *Ms*PimA•GDP-Man, *Ms*PimA•Man-P and *Ms*PimA•Man-

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PP-RIB showed remarkable different $T_{\rm M}$ values of 44.3, 36.5 °C and 36.2 °C (Table 1). The addition of GDP-Man significantly increased the Δ H to 165 kcal/mol, whereas the addition of Man-P or Man-PP-RIB had no effect on *Ms*PimA Δ H parameters (Table 1).

The 'active extended' to 'active compact' transition of MsPimA as visualized by SAXS

Small angle X-ray scattering (SAXS) has proved to be a powerful technique capable of providing structural information of flexible and dynamic proteins in solution (15, 19). The interatomic distance distribution function (P(r)) computed for MsPimA•GDP-Man, MsPimA•GMP, MsPimA•PP-RIB, MsPimA•Man-P, and MsPimA•Man-PP-RIB is shown in Figure 5A. (see also Table S2). The molecular mass determined from the scattering data confirmed that the protein is monomeric in solution in agreement with size exclusion chromatography and analytical ultracentrifugation (11, 18). The radius of gyration (R_g) value obtained for MsPimA•GDP-Man (29.3(5) Å) and the unliganded form (30.8(4) Å), revealed a reduction in Rg (ΔR_g) of -1.5 Å, indicating that the complex is in a compact conformation with respect to MsPimA alone, a similar behavior observed for the MsPimA•GDP complex (29.2(4) Å; (ΔR_g of -1.6 Å). Interestingly, the addition of GMP ($R_g = 29.6(3)$ Å) also trigerred the formation of a compact conformation as visualized in the crystal structure of the MsPimA•GMP complex (Figure 2). In contrast, the addition of PP-RIB ($R_g = 31.1(4)$ Å), Man-P ($R_g = 31.0(5)$ Å) or Man-PP-RIB ($R_g = 31.2(4)$ Å) to MsPimA showed similar R_g values as for the unliganded form of the enzyme ($R_g = 30.8(4)$ Å; 21). Furthermore, the normalized Kratky plot (Figure 5B) indicates that MsPimA in complex with GDP-Man, GDP or GMP is more symmetric (closer to following Guinier's approximation), than MsPimA unliganded or in complex with PP-RIB, Man-P or Man-PP-RIB. Normalized P(r) function distributions are shown in Figure 5C.





Figure 5. The active extended to active compact transition as visualized by SAXS. A. Scattering curves of unliganded *Ms*PimA (black), *Ms*PimA•GDP-Man (blue), *Ms*PimA•GDP (red), *Ms*PimA•GMP (light blue), *Ms*PimA•Man-PP-Rib (pink), *Ms*PimA•PP-Rib (violet) and PimA•Man-P (green) complexes. *B.* Normalized Kratky plot of unliganded *Ms*PimA (black), *Ms*PimA•GDP-Man (blue), *Ms*PimA•GDP (red), *Ms*PimA•GMP (light blue), *Ms*PimA•GDP-Rib (pink), *Ms*PimA•GDP-Man (blue), *Ms*PimA•GDP (red), *Ms*PimA•GMP (light blue), *Ms*PimA•Man-PP-Rib (pink), *Ms*PimA•GDP-Rib (violet) and PimA•Man-P (green) complexes. Lines crossing at $q * R_g = \sqrt{3}$ with a magnitude of 1.104 indicates the expected maxima of a globlular particle obeying Guinier's approximation. *C.* Normalized *P*(r) function distributions.

DISCUSSION

The crystal structures of the *Ms*PimA•GDP-Man and *Ms*PimA•GDP complexes were obtained in an 'active compact' conformation (Figure 1) (11). Interestingly, the MsPimA•GMP complex also revealed an active compact conformation, but with important differences on the structural arrangement of the GMP molecule (Figure 2) in the active site when compared with MsPimA•GDP-Man and MsPimA•GDP complexes (Figure 1). This reorganization changes the previously seen interactions with the β -PO₄ substantially and instead reveals new ones with the α -PO₄. From a catalytic point of view, the reaction catalyzed by PimA involves the cleavage of the phosphor sugar bond. Bearing this in mind, the different α -PO₄ disposition within the active site can be seen as an attempt to accommodate the GMP molecule and to stabilize temporary and weakly key interactions originally needed for the β -PO₄ cleavage. Actually, SAXS analysis of the corresponding MsPimA•GDP-Man, MsPimA•GDP and MsPimA•GMP complexes are in agreement with the ability of GDP-Man, GDP and GMP to stabilize the 'active compact' conformation (Figure 5). It is also observed by limited proteolysis, where *Ms*PimA is less prone to protease cleavage when bound to GDP-Man, GDP and GMP (Figure 3), and by a substrate-complexity proportional gain in tertiary structure by near-UV spectra (Figure 4). Nevertheless, this gain is not accompanied by an increase in the van't Hoff enthalpy suggesting no stabilization. Experimentally calculated $T_{\rm M}$ from our CD assays are in agreement with this observation, since only GDP-Man and GDP markedly stabilized MsPimA by ca. 10°C and the derivatives including GMP, GNO, GUA did not (Figure 4). The stabilization and binding are supported by the K_d parameters determined by ITC which revealed that GDP-Man and GDP bind to MsPimA in enthalpydriven reactions with dissociation constants, K_d in the micromolar range (11) whereas no binding of GMP, GNO, GUA could be detected ($K_d > 100 \mu$ M) (Figure 6).

The GMP structural data – crystal structure and SAXS, limited proteolysis and Near-UV spectra serve as evidence to propose that the nucleotide moiety linked at least to the α -PO₄ is able to weakly bind and temporarily stabilize the active conformation but, apparently it does not provide the sufficient energy to form a high affinity binding. Despite only a weak binding/stabilization is observed, the effect is still detectable by SAXS, limited proteolysis, and near-UV CD spectra. The GMP-bound form would also reflect the expected rigidity needed for the active sites in order to stabilize ligand-enzyme interactions (7, 8). Therefore, we propose the 'nucleoside moiety + α -PO₄ + β -PO₄' as minimal structure required to stabilize this conformation. Hence, the GMP effect on PimA might be seen as a sort of artifact due to the structural analogy with the native substrate. The essential interactions directly involved in the binding to β -PO₄, mainly arising from Gly16, Arg196, and Lys202, are distorted when GMP accommodates within the active site (Figure 2). In addition, the disposition of the catalytic loop containing Ser275 and Glu274 is also altered. As our biochemical data showed, it is reasonable to propose that these residues and their interaction with the β -PO₄ confer the specificity for the high affinity binding and the stabilization energy to achieve a competent active site in the active compact conformation. Additionally, we also studied the contribution of the mannose moiety and its derivatives. SAXS data for MsPimA•Man-P, MsPimA•Man-PP and MsPimA•PP-RIB, complexes also displayed similar behavior to the unliganded form. Our CD, limited proteolysis and ITC experiments performed in the presence of the above mentioned truncated versions are in agreement with our hypothesis showing no thermal stabilization in the presence of PP-RIB, Man-PP-RIB, and Man-P nor binding detected to MsPimA ($K_d > 100 \mu$ M) (Figure 6) and in consequence, they were more prone to the protease's activity (Figure 3). In conclusion, only GDP-Man and GDP exhibit high affinity binding constants and the highest stabilization energies to interact with the active compact state. Therefore, the simultaneous presence of the nucleotide moiety and

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the β -PO₄ moieties is indispensable to (i) provide the high affinity binding and (ii) to generate a competent active site through the active compact conformation stabilization. Nevertheless, the nucleotide moiety bound to at least the α -PO₄ is able to interact weakly with PimA favoring the active compact conformation state due to a delocalization of the α -PO₄, almost reaching the β -PO₄ location, and establishing a mimicking interaction.



Figure 6. Graph showing the R_g , K_d and T_M values obtained for PimA in the presence or absence of different GDP-Man derivatives. A. Rg Guinier (filled black circles) and Rg real space (filled red squares) values of unliganded MsPimA, MsPimA•GDP-Man, MsPimA•GDP, MsPimA•GMP, MsPimA•GNO, MsPimA•GUA, MsPimA•Man-PP-Rib, MsPimA•PP-Rib and MsPimA•Man-P complexes are ploted with reference to the vertical left- and right-axis, respectively. The K_d values are indicated next to the ploted Rg values. B. T_M (empty circles) values of unliganded MsPimA, MsPimA•GDP-Man, MsPimA•GDP, MsPimA•GMP, MsPimA•GNO, MsPimA•GUA, MsPimA•Man-PP-Rib, MsPimA•PP-Rib and MsPimA•Man-P complexes are ploted with reference to the vertical leftand right-axis, respectively. The K_d values are indicated next to the ploted T_M values.

Two catalytic mechanisms have been proposed for retaining GTs, (i) a double displacement mechanism that involves a nucleophilic residue that forms a covalent sugarenzyme intermediate, and in the absence of a nucleophilic residue, (ii) a substrate-assisted front-face reaction mechanism that involves the formation of an oxocarbenium ion-like transition state (14, 31, 32, 33). In support of the substrate-assisted front-face reaction mechanism, recent structural work on GT-A GTs using a quick-soaking method has provided unprecedented insights by trapping snapshots along the reaction cycle (33, 34, 35). That is the

case for the retaining GTs glucosyl-3-phosphoglycerate synthase (GpgS) and α -1,3galactosyltransferase (α 3GalT). Although the latter contains a putative nucleophile in the active site (Glu317), the native ternary complexes determined in a productive mode for catalysis in the presence of their sugar donor, acceptor substrates, and metal cofactors, support the substrate-assisted front-face reaction mechanism, suggesting a common reaction mechanism for all retaining GTs (34, 36, 37).

PimA and PimB are two GT-B GT that exhibit a putative nucleophile in the active site (residues Glu274 and Glu290, respectively). But, although the mutation Glu274Ala in PimA completely abolished the enzyme's activity (18), the Glu290Ala mutation in PimB did not reduce activity to the same extent (38), suggesting that this conserved glutamate residue might be important for sugar binding and stabilization of the transition state. Thus, it has been proposed that PimA is likely to follow a substrate assisted front-face reaction mechanism (39). When the position of the NDP-sugar donor within the active site was compared with closest homologs WaaG and MurG in the presence of UDP-2-deoxy-2-fluoroglucose (UDP-2FGlc), the pyrophosphate-sugar moiety in the PimA complex has a different conformation compared to WaaG and MurG. Nevertheless, GDP-Man is found in a 'bend back' conformation relative to the pyrophosphate group and sugar residues which is conserved in all retaining GTs, exposing the anomeric C1 carbon to nucleophilic attack (40). Unfortunately, there is no structural information of PimA in the presence of the acceptor PI, but QM/MM calculations for two of the closest structural and functional homologs, trehalose-6-phosphate synthase (OtsA) and MshA, support a substrate-assisted front-face reaction mechanism (31, 41).

OtsA is an extensively studied retaining GT-B that uses UDP-Glucose as donor and Glc-6P as acceptor to form trehalose-6P. The crystal structure of OtsA in complex with the glycomimetic inhibitor validoxylamine A 6'-O-phosphate and UDP traped a putative

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transition state–like arrangement in the active site (42). This structure provided structural evidences for a front-face reaction mechanism, which was further supported by kinetic isotope effects that indicated the existence of a highly dissociative oxocarbenium ion–like transition state (31). These experimental obsrvations are in agreement with QM/MM calculations, which suggest that the reaction is likely to happen stepwise and that it involves the formation a short-lived oxocarbenium-ion-like intermediate. These calculations suggested the phosphate group acts as a base enhancing the nucleophilicity of the incoming acceptor molecule (32).

Another example, the GT4 MshA from Corynebacterium glutamicum involved in the mycothiol biosynthesis, illustrates a very similar scenario to that possibly undergone by PimA. MshA is a metal-independent GT-B which uses 1-L-myo-inositol-1-phosphate (m-Ins-1P), and UDP-GlcNAc as activated donor, transferring the GlcNAc to the 3rd position of the inositol ring (40). In this case, a concerted substrate-assisted front-face reaction mechanism has been proposed where the m-Ins-1P O3 performs a nucleophilic attack on the anomeric C1 carbon in GlcNAc. As opposed to the stepwise front-face reaction mechanism of OtsA, here the m-Ins-1P O3 is concomitantly deprotonated by the oxygen 3 of the β -PO₄ (O3B) group that acts as leaving group. The key process leading to the transition state clearly corresponds to the breaking of the GlcNAc – β -PO₄ bond with the concomitant strengthening of the hydrogen bond between the β -PO₄ 3B and the HO3 of the m-Ins-P (41). A close inspection of the interaction between the MshA residues and the β -PO₄ reveals that MshA, similarly to PimA, has a glycine, Gly23, (equivalent to Gly16 in PimA) coordinating the pyrophosphate. The Gly16, which belongs to the named GlyGly loop - $\beta 1/\alpha 1$ loop located at the N-terminal domain - is conserved in other GT-B superfamily members and its motion appears to be essential for catalysis in the closest structural homologs (37, 43). In addition, Arg231 in MshA (equivalent Arg196 in PimA) is stabilizing and placing β -PO₄ in close contact with the

 O3 nucleophile from the donor. The mutations of Arg196Ala in PimA completely abolished the activity (11) indicating the relevance of this residue for catalysis. Our structural results, the comparison with retaining GT-B homologs and computational studies support a front-face reaction mechanism that involves the formation of an oxocarbenium ion-like transition state, where the β -PO₄ must be placed spatially close to the acceptor group in order to assist in catalysis. The residues responsible of the proper cleavage and spatial location of β -PO₄ are crucial for catalysis corroborating observations that the β -PO₄ is essential. Nonetheless, further structural and computational evidence are required for PimA.

It is well established that for the GT-B superfamily members the acceptor binding site is located at the N-terminal domain whereas the donor substrate binds mainly to the C-terminal domain. The catalytic cleft is formed between the two spatially separated domains upon an 'open-to-closed' conformational change. This domain rearrangement has been extensively characterized for PimA and predicted and/or observed for some other members of the GT-B superfamily (14, 46, 47, 48). PimA is therefore an outstanding example of GT-B flexibility and intrinsic dynamic properties for the study of GT-B enzymes. The coexistence of different substates in solution serves as evidence to show how PimA samples, at least, three different conformations. These motions are mainly happening at the N-terminal domain of the protein unveiling the potential relevance of the acceptor substrate impact on conformational changes. Since the range of acceptors is wider than the range of donors, in terms of chemical structure complexity, this domain exhibits a greater diversity and flexibility of secondary structure elements arrangements (37, 49) which in turn, facilitates the adaptability to interact with acceptor substrates. However, despite the apparent structural disparity in the N-terminal domain for some GT-B members, evolution has driven the GT-B function based on (i) conserved motifs responsible for the recognition and binding to the substrates and (ii) high intrinsic flexibility presenting a shared trend in dynamics – 'open to closed' motion. The

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open-to-closed movement, as a common feature, indicates that it is efficient and functional motion required to perform the catalysis as it places donor and acceptor substrates in close spatial contact (37, 43).

For PimA, the scenario seems to be even more complex due to a fold switch prior to the 'open-to-closed' motion. The fold switch is considered the rate-limiting step of the glycosylation reaction and it has been defined as the activation mechanism of PimA (15). Once the fold switch occurs, it is expected that PimA populates an active extended structurally This conformation, not-yet characterized. domain 'open-to-closed' conformational change allows critical residues from the N- and C-terminal domains to bring together, and build a functionally competent active site (Figure 1). Up to the date, it was observed that the donor substrate presence induced the 'open-to-closed' motion in some members (13, 37). However, the dynamic process and the structural determinants responsible for it remained unresolved. As we have shown, the residues intervening in the β -PO₄ cleavage are critical to stabilizing this state and to the catalysis. Indeed, our results are in agreement with the evidence for a correlation between an increase in the structural complexity of the substrate and the derived conformational changes since they are stabilized by the increasing number of key side chain interactions while the substrate binding provides the energy to form the transition state (11). Conversely, a decrease in the donor substrate structure complexity would lead to a lower binding and stabilization interactions. For PimA, this agrees until certain minimal structure of the donor substrate. We have demonstrated that the interactions with the nucleotide moiety and the β -PO₄ are important to stabilize the active compact conformation in which the catalysis occurs. The intrinsic flexibility of PimA might be finely tuned and guided by the donor substrate structural determinants having an impact ultimately in its catalytic activity rate. This study provides an important contribution toward the understanding of such conformational transition, an activation mechanism, in the framework of glycosyl transfer reactions mediated by GT-B enzymes.

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CONFLICT OF INTEREST STATEMENT

None declared.

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