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Ultra-Fast Light-Driven Electron Transfer in a Ru(II)tris(bipyridine)-Labelled Multiheme Cytochrome

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Abstract:

Multiheme cytochromes attract much attention for their electron transport properties. These proteins conduct electrons across bacterial cell walls, along extracellular filaments, and when purified can serve as bionanoelectronic junctions. Thus, it is important and necessary to identify and understand the factors governing electron transfer in this family of proteins. To this end we have used ultra-fast transient absorbance spectroscopy, to define heme-heme electron transfer dynamics in the representative multiheme cytochrome STC from *Shewanella oneidensis* in aqueous solution. STC was photo-sensitized by site-selective labelling with a Ru(II)(bipyridine)₃ dye and the dynamics of light-driven electron transfer described by a kinetic model corroborated by molecular dynamics simulation and density functional theory calculations. With the dye attached adjacent to STC Heme IV, a rate constant of $87 \times 10^6 \text{ s}^{-1}$ was resolved for Heme IV \rightarrow Heme III electron transfer. With the dye attached adjacent to STC Heme I, at the opposite terminus of the tetraheme chain, a rate constant of $125 \times 10^6 \text{ s}^{-1}$ was defined for Heme I \rightarrow Heme II electron transfer. These rates are an order of magnitude faster than previously computed values for unlabeled STC. The Heme III/IV and I/II pairs exemplify the T-shaped heme packing arrangement, prevalent in multiheme cytochromes, whereby the adjacent porphyrin rings lie at 90° with edge-edge (Fe-Fe) distances of ≈ 6 (11) Å. The results are significant in demonstrating the opportunities for pump-probe spectroscopies to resolve inter-heme electron transfer in Ru-labeled multiheme cytochromes.

Introduction:

Species of *Shewanella* attract much interest for their ability to respire in the absence of oxygen by transferring electrons from intracellular oxidation of organic matter to extracellular acceptors including Fe₂O₃ and MnO₂ nanoparticles.¹⁻² Multiheme cytochromes are essential to this process and these fascinating proteins are spanned by chains of close-packed *c*-type hemes. Intra- and inter-cytochrome electron transfer occurs by complementary Fe(III) \leftrightarrow Fe(II) transitions of neighboring sites³⁻⁵ and in this way electrons are moved from the inner bacterial membrane, across the periplasm and outer membrane lipid bilayer to reach the cell exterior. Multiheme cytochromes also contribute to the conductivity of extracellular structures, often termed bacterial nanowires, which transfer electrons across distances greatly exceeding cellular dimensions. These structures for *Shewanella oneidensis* are multiheme cytochrome containing extensions of the bacterial outer membrane⁶ and for *Geobacter sulfurreducens* are filaments⁷⁻⁸ comprised of a polymerized multiheme cytochrome. Beyond their biological role, the remarkable electron transfer properties of multiheme cytochromes have stimulated interest in these proteins as novel bioelectronic junctions and devices.⁹⁻¹² Furthermore, these proteins underpin the wiring of bacteria to electrodes^{1, 13-15} to produce electricity in mediator-less microbial fuel cells and valued chemicals by microbial

electrosynthesis. It is now important to identify the factors governing electron transfer in this family of proteins to both understand biology and inspire advances in new, and yet to be conceived, biotechnology.

Multiheme cytochromes are defined by the presence of close-packed *c*-type hemes, typically with His/His axial ligation, arranged in similar configurations despite very different amino acid sequences and protein folds.³ Two heme-packing motifs, namely T-shaped and stacked, predominate in the structures resolved to date. Both motifs are present in the periplasmic cytochrome STC¹⁶ from *S. oneidensis* that is spanned by a chain of four His/His ligated hemes, Fig. 1. Heme pairs I/II and III/IV exemplify the T-shaped geometry of neighbors with perpendicular porphyrin rings and edge-edge (Fe-Fe) distances of 5-8 (11-12) Å. The STC heme II/III pair exemplifies the stacked packing motif with parallel porphyrin rings in van der Waals contact and a shorter edge-edge (Fe-Fe) distance of ≈ 4 (≈ 9) Å. The possibility that these geometries are optimized to impose control over electron transfer rates and direction has been explored at a single-protein level through quantum chemistry and molecular simulation.¹⁷⁻²¹ However, to the best of our knowledge, direct measurements of heme-heme electron transfer rates have yet to be reported for STC, or other multiheme cytochromes. As a consequence we were motivated to establish whether pump-probe spectroscopy could provide experimental insight into STC heme-heme electron transfer dynamics and, in turn, inform discussions surrounding the mechanism of electron transfer in multiheme cytochromes.

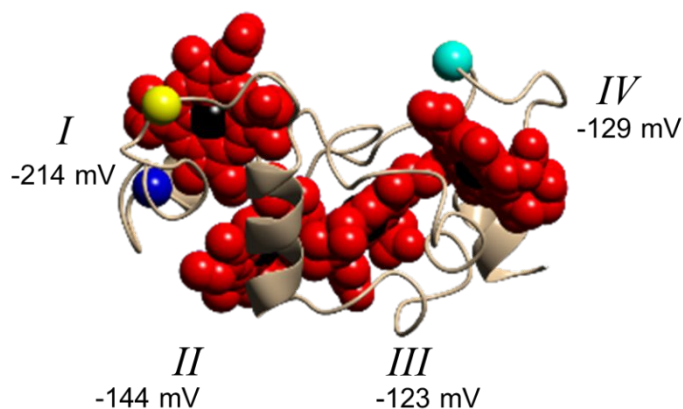
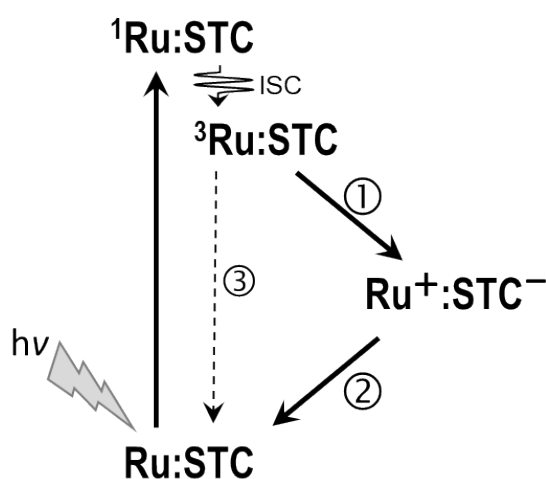


Figure 1. STC of *S. oneidensis* illustrating the four hemes (red) and their microscopic reduction potentials²⁸ in the oxidized protein. The C $_{\alpha}$ atoms of residues changed to Cys for labelling with a Ru(II)-dye photosensitizer are indicated as circles: residues 10 (blue), 23 (yellow) and 77 (cyan).

Pump-probe spectroscopies, through appropriate combinations of light-triggered electron transfer and time-resolved spectroscopy, offer a powerful way to resolve pathways and dynamics of protein electron transfer across timescales ranging from pico- to milli-seconds.²²⁻²⁵ The heme-heme electron transfer rate constants in solvated

STC are calculated¹⁸ to range from $\approx 0.5\text{--}200 \times 10^6 \text{ s}^{-1}$ and in previous work⁹ we established that STC could be labelled site-selectively with $[\text{Ru(II)(4-bromomethyl-4'-methylbipyridine) (bipyridine)}_2]^{2+}$, a thiol-reactive phototrigger of electron transfer.^{22, 26-27} Following photoexcitation into the Ru-dye metal-to-ligand charge transfer (MLCT) band, the triplet excited state was oxidatively quenched by heme reduction.⁹ Such electron transfer, Scheme 1 ① ($E_m \text{ Ru(III)/(II}^*) \approx -870 \text{ mV vs SHE}^{26}$), produces a charge separated state, $\text{Ru}^+:\text{STC}^-$, that will return to the $\text{Ru}:\text{STC}$ ground state by charge recombination, Scheme 1 ② ($E_m \text{ Ru(III)/(II)} \approx 1270 \text{ mV}^{26}$). However, heme-heme electron transfer in $\text{Ru}^+:\text{STC}^-$ could result in each heme existing transiently as Fe(II) ; the corresponding microscopic E_m values²⁸ lie between -120 and -215 mV as summarized in Figure 1. Gaining direct spectroscopic evidence for electron transfer along the heme wire will be challenging due to the chemical similarity of the hemes. However, we reasoned that heme-heme electron transfer will influence the dynamics of the corresponding photocycle in a manner that could be resolved by ultra-fast pump-probe spectroscopy given the timescales predicted for heme-heme electron transfer and lack of protein superstructure, that will inevitably place the Ru(II) -dye in close proximity to the acceptor heme leading to fast charge separation and recombination rates.

Here we present ultra-fast transient absorbance (TA) of STC proteins photosensitized to inject an electron into opposite ends of the tetraheme chain, into either Heme I or Heme IV as illustrated in Figure 1. Kinetic modelling of the electron transfer dynamics, corroborated by molecular dynamics (MD) simulations and density functional theory (DFT) calculations that provide a microscopic view of the contributing processes, allows us to present rate constants for $\text{Heme I} \leftrightarrow \text{Heme II}$ and $\text{Heme IV} \leftrightarrow \text{Heme III}$ electron transfer that are indicative of fast heme-to-heme electron transfer on the 10 ns time scale. The results are significant in demonstrating the opportunities for pump-probe spectroscopies to resolve inter-heme electron transfer in Ru-labeled multiheme cytochromes.



Scheme 1

Experimental Methods:

Sample Details: Ru(II)(4-bromomethyl-4'-methylbipyridine) (bipyridine)₂(PF₆)₂ (HetCat Switzerland) was prepared as previously described.²⁷ All other reagents were analytical grade and aqueous solutions prepared with water having resistivity > 18 MΩ cm. Preparation of the STC variants A10C, T23C and S77C and of their photosensitized forms, here termed Ru₁₀:STC, Ru₂₃:STC and Ru₇₇:STC respectively, was as previously described⁹ and outlined in the Supporting Information. Protein concentrations were defined by electronic absorbance of the tetra-Fe(III) forms using $\epsilon_{407\text{ nm}} = 422\text{ mM}^{-1}\text{ cm}^{-1}$ or $\epsilon_{552\text{ nm}} = 29.1\text{ mM}^{-1}\text{ cm}^{-1}$ as reported by Leys *et al.*¹⁶ TA measurements were performed with anaerobic solutions containing 20 mM TRIS-HCl, 0.1 M NaCl at pH 8.5 and in the absence of sacrificial redox partners. All hemes were in the oxidized, i.e. Fe(III) state, prior to irradiation. Measurements were performed at two protein concentrations to ensure equally good signal-to-noise ratios for quantitative analysis at each wavelength of interest; $\approx 20\text{ }\mu\text{M}$ protein for 369 and 419 nm, and $\approx 160\text{ }\mu\text{M}$ protein for the less intense features at 453 and 552 nm. The weight average molecular mass and oligomeric state of Ru₇₇:STC in solutions of 20 mM TRIS-HCl, 0.1 M NaCl at pH 8.5 was defined by analytical ultracentrifugation, sedimentation equilibrium analysis, using a Beckman Optima XL-1 analytical ultracentrifuge equipped with scanning absorbance optics and a Ti50 rotor. Analytical gel filtration was performed with a Superose 6 Increase 10/300 column (GE Healthcare).

Time-Resolved Multiple-Probe Spectroscopy (TRMPS): TRMPS TA was performed at the Central Laser Facility of the Rutherford Appleton Laboratory using the apparatus described previously.²⁹⁻³⁰ Excitation was at 457 nm with 100 fs pulses of 400 nJ at 1 kHz focused to a spot size of 200 μm FWHM. The experiment employed two synchronized ultrafast lasers, one operating at 1 kHz for excitation and one at 10 kHz (the probe), arranged such that for every pump pulse ten probe spectra are collected and pump-probe time delays between 200 fs and 1 ms are accessible. TA difference spectra were recorded using a white light continuum generated in CaF₂, with alternate pump pulses blocked by a chopper. Samples were rastered in the beam to ensure a fresh spot was irradiated by each pump pulse. Further details of the data collection and full details of the data processing are provided in the Supporting Information.

Model Fitting for Electron Transfer Dynamics: As explained in *Results*, the TA signals were fit to two types of kinetic models shown in Scheme 2 and Schemes 3 and 4. The time-dependent populations for the species in Scheme 2, where y indicates the Ru-dye attachment site, are given by:

$$p(^3\text{Ru}_y:\text{STC}) = \sum_x p_{0,x} \cdot e^{-k^x_{\text{CS}}t}, \quad (\text{Eq. 1})$$

$$p(\text{Ru}_y^+:\text{STC}^-) = \sum_x p_{0,x} \cdot k^x_{\text{CS}} \cdot (e^{-k^x_{\text{CS}}t} - e^{-k^x_{\text{CR}}t}) / (k^x_{\text{CR}} - k^x_{\text{CS}}), \quad (\text{Eq. 2})$$

$$p(\text{Ru}_{77}:\text{STC}) = \sum_x p_{0,x} \cdot (1 + (k_{\text{CS}}^x \cdot e^{-k_{\text{CR}}^x t} - k_{\text{CR}}^x \cdot e^{-k_{\text{CS}}^x t}) / (k_{\text{CR}}^x - k_{\text{CS}}^x)), \quad (\text{Eq. 3})$$

and $\sum_x p_{0,x} = 1$, where $p_{0,x}$ is the initial triplet state concentration of $\text{Ru}_{77}:\text{STC}$ conformer x . Three such models were considered comprised of 2 ($x = \mathbf{a}, \mathbf{b}$), 3 ($x = \mathbf{a}, \mathbf{b}, \mathbf{c}$) and 4 ($x = \mathbf{a}, \mathbf{b}, \mathbf{c}, \mathbf{d}$) distinct conformers. For each model the set of rate constants k_{CS}^x , k_{CR}^x and the initial concentrations of each conformer $p_{0,x}$ were determined from best simultaneous fits of the experimental populations to Eqs. (1)-(2), see Supporting Information for details on the fitting procedure. With regard to the refined kinetic model of Scheme 3 including electron transfer from Heme IV to Heme III of $\text{Ru}_{77}:\text{STC}$, the set of kinetic rate equations Eqs. (4)-(7),

$$d(p(^3\text{Ru}_{77}:\text{STC}_x))/dt = -k_{\text{CS}}^x \cdot p(^3\text{Ru}_{77}:\text{STC}_x) \quad (\text{Eq. 4})$$

$$d(p(\text{CS}_x^{\text{IV}}))/dt = k_{\text{CS}}^x \cdot p(^3\text{Ru}_{77}:\text{STC}_x) + k_{\text{IV,III}} \cdot p(\text{CS}_x^{\text{III}}) - (k_{\text{III,IV}} + k_{\text{CR}}^x) \cdot p(\text{CS}_x^{\text{IV}}) \quad (\text{Eq. 5})$$

$$d(p(\text{CS}_x^{\text{III}}))/dt = k_{\text{III,IV}} \cdot p(\text{CS}_x^{\text{IV}}) - k_{\text{IV,III}} \cdot p(\text{CS}_x^{\text{III}}) \quad (\text{Eq. 6})$$

$$d(p(\text{Ru}_{77}:\text{STC}_x))/dt = k_{\text{CR}}^x \cdot p(\text{CS}_x^{\text{IV}}) \quad (\text{Eq. 7})$$

were solved analytically for a given set of rate constants and initial concentrations. The rate constant $k_{\text{IV,III}}$ was determined by $k_{\text{III,IV}}$ and the driving force for electron transfer assuming detailed balance, $k_{\text{IV,III}} = k_{\text{III,IV}} \cdot e^{\frac{F(\epsilon_{\text{III}}(\text{O}) - \epsilon_{\text{IV}}(\text{O}))}{k_{\text{B}}T}} = k_{\text{III,IV}} \cdot 0.79$ at $T = 300$ K, where $\epsilon_i(\text{O})$ are the reduction potentials of heme i in the all-oxidized (O) state of the protein, taken from experiment²⁸ and F is the Faraday constant. Hence, only one additional fitting parameter, $k_{\text{III,IV}}$, was added to the multiple conformer models of Scheme 2. The best set of fit parameters was determined using the same fitting procedure as for the models for Scheme 2. Equivalent equations and processes were used to account for Scheme 4 in providing descriptions of electron transfer between Hemes I and II of $\text{Ru}_{10}:\text{STC}$ and $\text{Ru}_{23}:\text{STC}$.

Atomistic Description of Electron Transfer Dynamics in $\text{Ru}_{77}:\text{STC}$. For docking of the Ru-dye, the structure¹⁶ of the native STC protein (1M1Q) was prepared in the all-oxidized state as described in our previous work¹⁸ except where indicated otherwise. All protein residues are in the standard protonation states at pH = 7 except one propionate from Heme III, which is protonated. Serine residue 77 was replaced by a cysteine and the hydrogen atom of S-H by $\text{Ru}(\text{bpy})_2(4\text{-CH}_2\text{-4'-methylbipyridine})$. The docking of the label was performed by sampling the three dihedral angles τ_1, τ_2, τ_3 , defined in Fig. 7A, between 0° to 360° in increments of 5°. In this way more than 100,000 trial structures were generated. All structures were energy minimized and clustered with respect to the dihedral angles, see Fig. S9, resulting in a total of four unique low-energy clusters termed in the following as conformers **1**, **2**, **3** and **4**. These four conformers were used as initial structures for MD simulations employing the AMBER03 protein force field³¹, TIP3P water³² and the force field parameters for the Ru-label from our

previous work.³³ The conformers were solvated with a shell of 15 Å of water with Na⁺ and Cl⁻ counter ions added to correspond to an ionic strength of 0.1 M. After equilibration in the ground electronic state (Ru²⁺, all hemes Fe³⁺) to 300 K and 1.013 bar (MD time step = 2 fs), MD production runs were carried out for 100 ns in the NPT ensemble, for each conformer. Equilibrated configurations from these runs were taken to initialize MD simulations in the CS^V state of Ru₇₇⁺:STC⁻ (Ru³⁺, Heme IV Fe²⁺, all other hemes Fe³⁺). After equilibration to 300 K and 1.013 bar, 20 ns MD production runs were carried out for each conformer. Average dihedrals and Ru label-Heme IV distances are summarized in Table S5.

Rate constants for charge separation and recombination were calculated using semiclassical electron transfer theory. Electronic coupling matrix elements were calculated using the fragment-orbital density functional theory (FODFT) method implementation³⁴⁻³⁵ in the CPMD package³⁶ and the PBE exchange correlation functional.³⁷ Coupling matrix elements for charge separation were calculated along the trajectories in state Ru₇₇:STC for each conformer. The triplet ³MLCT orbital³⁸ of the Ru(II)-label was selected to couple with the two (quasi-)degenerate frontier orbitals of Heme IV, which are composed of the Fe-d (*t*_{2g}) and heme ring orbitals. For charge recombination, couplings were calculated along the trajectories in state Ru⁺:STC⁻ between the two (quasi-)degenerate frontier orbitals of Heme IV and the highest occupied orbital of the label, composed of Ru-d (*t*_{2g})-bpy ring orbitals, see Fig. S12. The couplings for each of the two orbital pairs were square-averaged to an effective coupling matrix element using a similar approach to that previously described¹⁸⁻¹⁹ and scaled by a factor of 1.348 as recommended in the benchmark study.³⁹ The QM models used for these calculations are detailed in the Supporting Information. Reorganization free energies were obtained as half of the Stokes shift⁴⁰⁻⁴¹ obtained by calculating the vertical electron transfer energy along the trajectories in Ru₇₇:STC and Ru₇₇⁺:STC⁻ states for each conformer. The outer-sphere reorganization free energy was scaled as recommended¹⁷ to account for the missing electronic polarizability of this force field. Moreover, a non-ergodicity correction to the reorganization free energy was applied as recommended⁴² to account for the ultrafast time-scale of the experiments described here. To this end we applied the self-consistent iteration scheme suggested previously¹⁷, where the outer-sphere contributions that are slower than the actual electron transfer event are removed, see Fig. S11. The inner-sphere contribution was obtained from the usual 4-point scheme using DFT and the PBE functional. Driving forces were taken from experiment.^{26, 28} All electron transfer parameters and rate constants for the four conformers are summarized in Table S6. Full details on docking, MD simulations, and calculation of electron transfer parameters can be found in the Supporting Information.

Results:

In the following sections we first present a detailed account of our studies of Ru₇₇:STC where the Ru-dye lies adjacent to Heme IV, Fig. 1. Features in the TA are assigned to states within the photocycle of Scheme 1 and their

transient populations accurately reproduced by a kinetic model that extends Scheme 1 to include electron transfer across the Heme IV \leftrightarrow Heme III pair. The kinetic model is validated by comparison to the predictions from atomistic descriptions of the protein and corresponding electron transfer dynamics. Finally TA of proteins having the Ru-dye adjacent to Heme I, Ru₂₃:STC and Ru₁₀:STC, is presented and interpreted through an equivalent model to define rate constants for electron transfer across the Heme I \leftrightarrow Heme II pair.

In the text below we use [Ru(bpy)₃]²⁺ to refer to the ground state of the thiol-reactive photosensitizer [Ru(II)(4-bromomethyl-4'-methylbipyridine)(bpy)₂]²⁺. The ground state of the photosensitizer attached to S77C STC, adjacent to Heme IV, is termed [Ru(bpy)₃]²⁺_{STC77} and an equivalent nomenclature describes the Ru-dye labelled STC proteins with cysteine at positions 10 and 23, adjacent to Heme I.

Photoexcitation of Ru₇₇:STC. The ground state absorbance of Ru₇₇:STC, Fig. 2A blue lines, is well described⁹ by superposition of the spectra for S77C STC and [Ru(bpy)₃]²⁺ in a 1:1 ratio. The dominant features arise from π - π^* transitions of the Fe(III) hemes; the maximum of the Soret-band occurs at 408 nm and of the α -/ β -band at 524 nm. The hemes also have significant absorbance at 452 nm where the MLCT band⁴³ of the Ru(II)-dye is centered, Fig. 2A red line. As a consequence, exciting the dye MLCT band by pulsed irradiation of Ru₇₇:STC at 457 nm also excites the hemes and the corresponding TA reports on the consequences of both processes. We extracted the TA associated with Scheme 1, arising solely from excitation of the Ru(II)-dye, through the following process described fully in the Supporting Information. The TA arising solely from electronically excited His/His ligated heme, e.g. Fig. S4, was defined by pulsed irradiation of S77C STC. The TA of S77C STC, after appropriate scaling to account for the concentration of excited heme in each sample, was then subtracted from the TA of Ru₇₇:STC to reveal the features associated with Scheme 1. The resulting differential spectra are presented in Fig. 3. We note that the photochemistry of Ru₇₇:STC extends to approx. 60 ns while that of S77C STC was complete within 50 ps, and that data points between 438 and 465 nm are excluded due to scattering of the pump pulse into the probe path.

TA Assignment and Analysis for Ru₇₇:STC. In the differential spectra of Fig. 3, chromophores with transiently depleted populations give negative features while those with transiently increased populations give positive features. The broad trough between \approx 400 and 500 nm describes depletion of the ground state dye concentration. At the shortest times, e.g. 0.7 ps, the positive features arise from absorbance by the dye triplet excited state, formally [Ru(III)(bpy)₂(bpy⁻)]²⁺_{STC77}. By comparison to the properties⁴³⁻⁴⁶ of analogous complexes, we attribute the peak at 369 nm to $\pi \rightarrow \pi^*$ transitions in the anionic bpy⁻ ligand and the broad positive feature above \approx 520 nm to the ligand-to-metal charge transfer (LMCT) band arising from neutral bpy-to-Ru(III) transitions. These assignments are supported by our TA, Fig. 4 and Fig. S3, of respectively [Ru(bpy)₃]²⁺ and the dye attached⁹ to the protein bovine serum albumin that contains no cofactors. We find no spectral evidence for the immediate

product of photoexcitation the singlet excited state, $^1\text{Ru}_{77}:\text{STC}$ Scheme 1, an observation in accord with the rapid intersystem crossing (ISC) reported⁴⁷⁻⁴⁸ for analogous systems.

Between ≈ 3 ps and 20 ns the differential spectra of Fig. 3 contain a narrow trough at 408 nm and peaks at 419, 522 and 552 nm. The Fe(III) hemes in $\text{Ru}_{77}:\text{STC}$ display a Soret maximum at 408 nm while the chemically reduced, tetra-Fe(II) forms display maxima at 419, 522 and 552 nm, Fig. 2A blue continuous line. Thus, the differential spectra reveal transient conversion of Fe(III) to Fe(II) heme and consequently formation of the $\text{Ru}_{77}^+:\text{STC}^-$ charge separated state. Direct spectral evidence for the oxidized dye, formally $[\text{Ru}(\text{III})(\text{bpy})_3]_{\text{STC}77}^{3+}$, is hard to discern. In analogous complexes the LMCT band⁴⁵ arising from the oxidized Ru-dye is a broad feature above ≈ 520 nm very similar in shape and intensity to that originating from the triplet excited state.

Having assigned all features in the differential spectra of Fig. 3 to species in Scheme 1, the most prominent feature(s) were used to define the transient populations of these species. Due to overlapping contributions from multiple chromophores this was facilitated by line shape fitting across an appropriate wavelength range as exemplified in Fig. 5 and fully described in the Supporting Information. For example, the absorbance due to $\text{Ru}_{77}^+:\text{STC}^-$ was defined at 419 nm, after removing contributions from the $\text{Ru}_{77}:\text{STC}$ ground state bleach and $\text{bpy}^- \pi-\pi^*$ absorption of $^3\text{Ru}_{77}:\text{STC}$, and independently at 552 nm, after removing contributions from the dye LMCT bands of $^3\text{Ru}_{77}:\text{STC}$ and $\text{Ru}_{77}^+:\text{STC}^-$. The results of both analyses are in good agreement, e.g. Fig. S5B. The $\text{Ru}_{77}^+:\text{STC}^-$ concentration was calculated from its absorbance at 419 nm using the differential extinction coefficient of $71\,427\text{ M}^{-1}\text{ cm}^{-1}$ obtained assuming each heme contributes equally to the absorbance of tetra-Fe(III) and tetra-Fe(II) STC, Fig 2B.

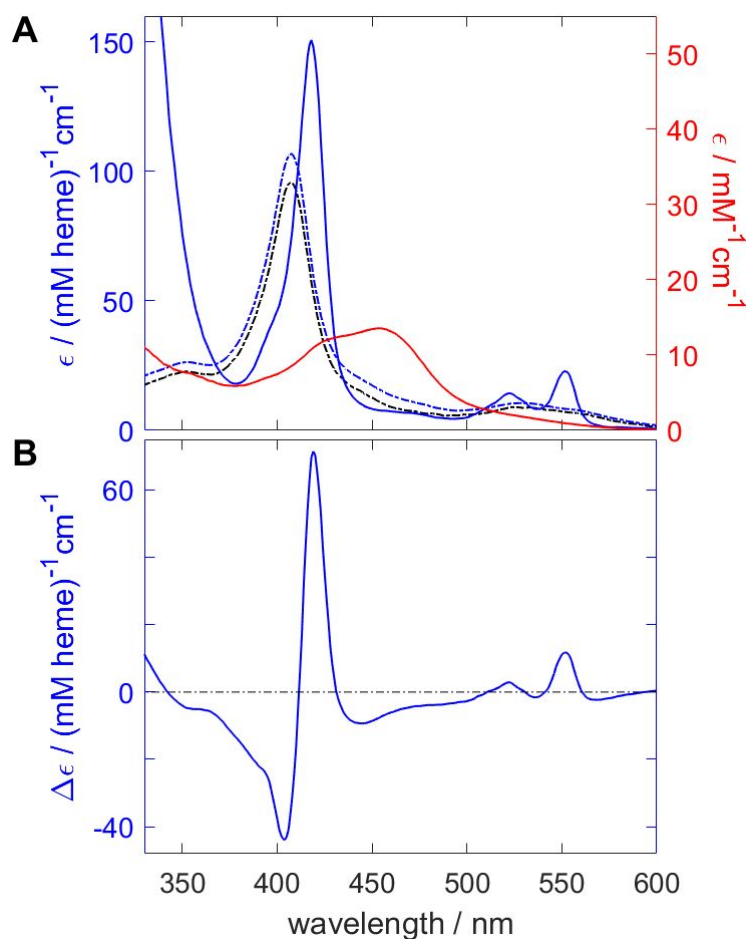


Figure 2. Ground state absorbance of $\text{Ru}_{77}\text{:STC}$ and $[\text{Ru}(\text{bpy})_3]^{2+}$

(A) Electronic absorbance of $\text{Ru}_{77}\text{:STC}$ in the oxidized, all Fe(III) state (blue broken line) and the dithionite reduced, all Fe(II) state (blue continuous line), together with that of oxidized S77C STC (black line) and $[\text{Ru}(\text{bpy})_3]^{2+}$ (red). Extinction coefficients for STC and $\text{Ru}\text{:STC}$ are derived from those reported by Leys *et al*¹⁶ assuming equal contribution from each heme. (B) Reduced *minus* oxidized difference spectrum for $\text{Ru}_{77}\text{:STC}$.

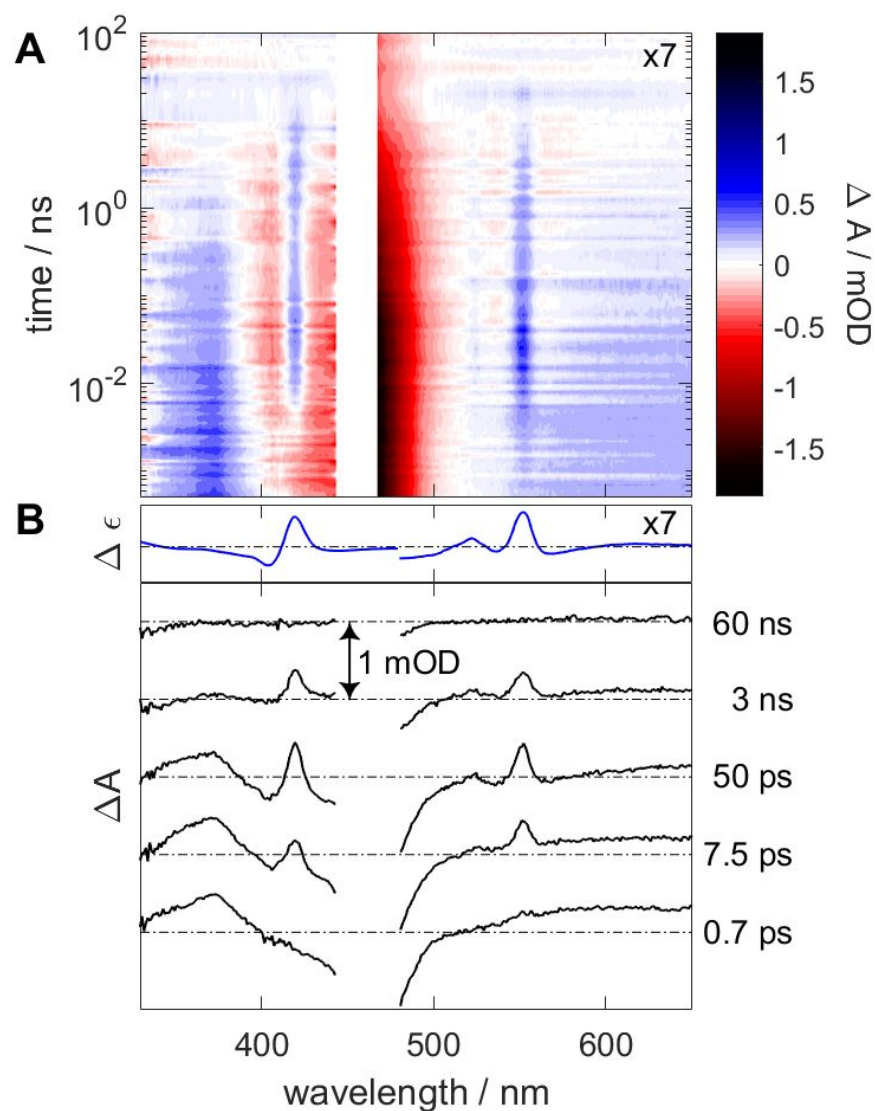


Figure 3. Transient absorbance of Ru₇₇:STC.

(A) Differential TA; blue = positive features, red = negative features. (B) TA spectra (black lines) for the indicated times after irradiation together with the difference spectrum (blue line) for chemically reduced *minus* oxidized Ru₇₇:STC with extinction coefficients omitted for clarity. Pulsed irradiation was at 457 nm and contributions from electronically excited hemes are removed, see text for details. Samples contained Ru₇₇:STC (22 μM for measurement < 440 nm, 160 μM for measurements > 480 nm) in anaerobic 20 mM Tris-HCl, NaCl, pH 8.5.

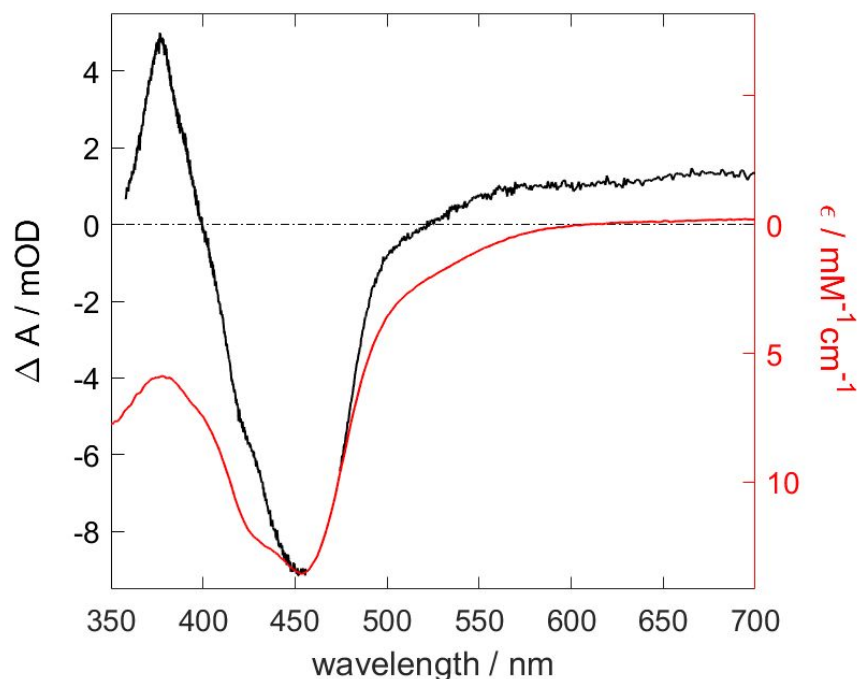


Figure 4. Spectral properties of $^3[\text{Ru}(\text{bpy})_3]^{2+}$ (black) and absorbance of $[\text{Ru}(\text{bpy})_3]^{2+}$ (red).

Differential TA 10 ps after excitation of the $[\text{Ru}(\text{bpy})_3]^{2+}$ dye (black line). The $[\text{Ru}(\text{bpy})_3]^{2+}$ ground state absorbance (red line) is presented with an inverted y-axis to best illustrate how ground state depletion contributes to the differential TA presented here and in Fig. 3.

The transient concentration of depleted $\text{Ru}_{77}:\text{STC}$ was defined by the dye ground state bleach using an extinction coefficient³⁸ of $14.6 \text{ mM}^{-1} \text{ cm}^{-1}$ for the dye MLCT band at 453 nm. The amplitude of the ground state bleach at this wavelength was derived by modelling with a Gaussian lineshape, e.g. Fig. 5. The transient concentration of $^3\text{Ru}_{77}:\text{STC}$ was obtained from the differential absorbance at 369 nm after accounting for contributions from the ground state bleach. For this wavelength, an extinction coefficient of $13.6 \text{ mM}^{-1} \text{ cm}^{-1}$ for $^3\text{Ru}_{77}:\text{STC}$ was derived from the TA of $[\text{Ru}(\text{bpy})_3]^{2+}$, Fig. 4. This is possible because after vibrational cooling, the dye ground state bleach describes a concentration equal to that of triplet state, and the latter species has negligible absorbance between 430 and 500 nm.⁴⁵

Figure 6 presents the outcomes of the above analysis, specifically, the transient populations of $^3\text{Ru}_{77}:\text{STC}$ and $\text{Ru}_{77}^+:\text{STC}^-$ together with recovery of the $\text{Ru}_{77}:\text{STC}$ population. No additional species are needed to describe the photocycle as the excited population is accounted for throughout.

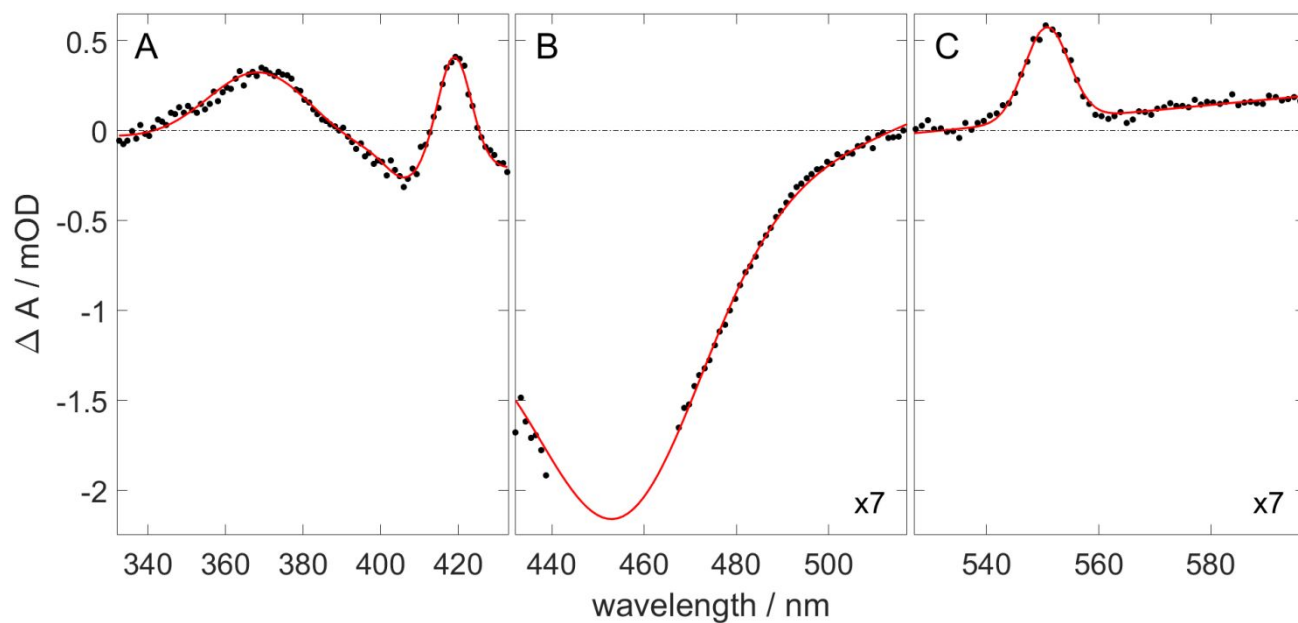


Figure 5. TA of Ru₇₇:STC at 30 ps after 457 nm excitation.

Data (black circles) and fits (red lines) for spectral windows covering (A) π - π^* absorption of the anionic bpy⁻ ligand in ³Ru₇₇:STC and the heme Soret band, (B) ground-state bleach of the Ru-dye, and, (C) heme α -band (552 nm maximum) and the Ru-dye LMCT band of ³Ru₇₇:STC and Ru₇₇⁺:STC⁻.

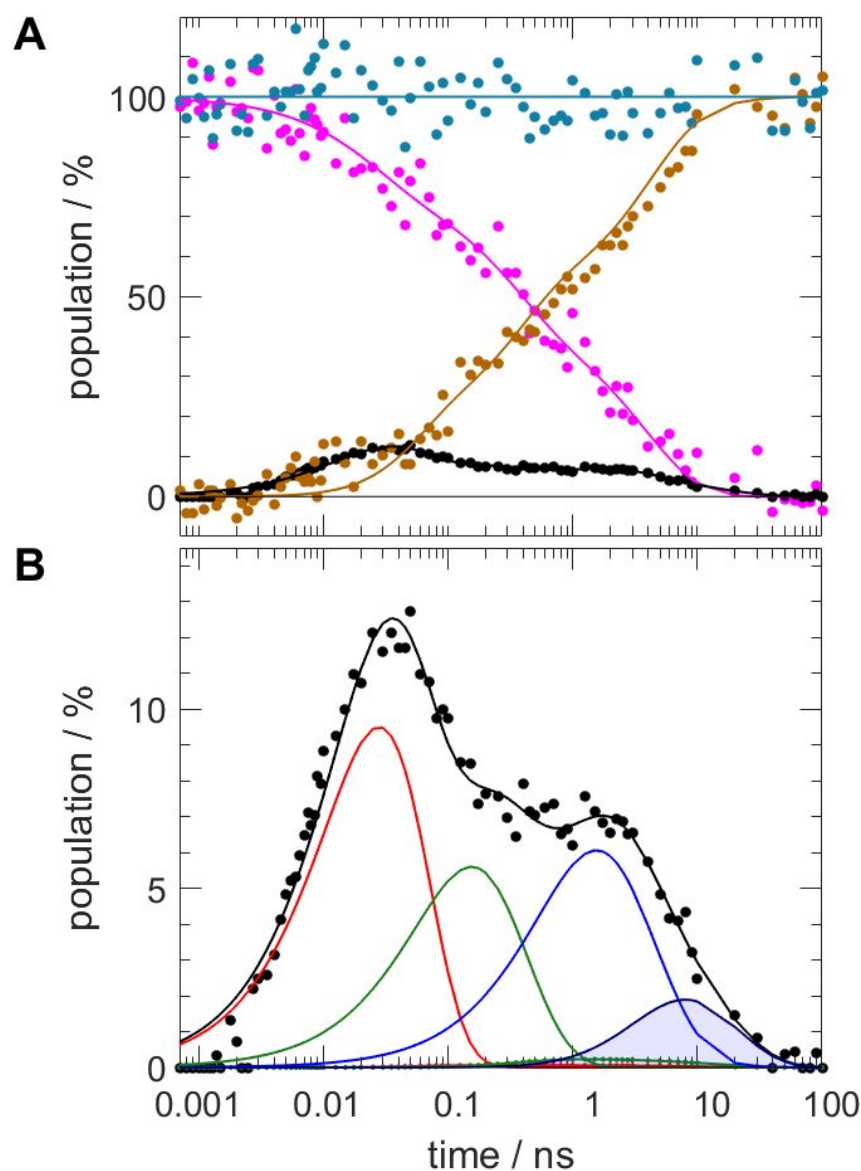
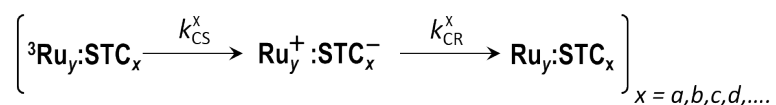


Figure 6. Progress of the Ru₇₇:STC Photocycle.

(A) Evolution of the experimentally defined concentrations (circles) of ³Ru₇₇:STC (pink) and Ru₇₇⁺:STC[−] (black) together with recovery of Ru₇₇:STC (brown) and the total concentration of these species (dark cyan). Fit to Scheme 3 (lines) with the parameters of Table 1 (bold). The population of ³Ru₇₇:STC at 0.7 ps was defined as 100%. (B) Expanded view of the population of Ru₇₇⁺:STC[−] shown in A) from experiment (black circles) and fit to Scheme 3 (black line) with the Table 1 parameters (bold). Contributions to the fit from CS^{IV} (lines, no fill) and CS^{III} (lines, shaded fill) are shown for conformers **a** (red), **b** (green) and **c** (blue). Note, contributions to CS^{III} from conformers **a** and **b** are very small and hardly visible.

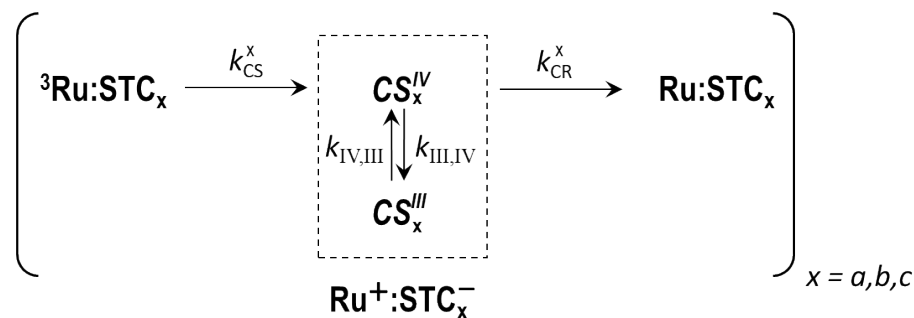
Electron Transfer Dynamics in Ru₇₇:STC. We have previously shown⁹ that the photoluminescence of the Ru-dye, Scheme 1 ③, is very significantly (> 98%) quenched when attached to S77C STC. Furthermore, in Ru₇₇:STC the charge separation, Scheme 1 ①, and charge recombination, Scheme 1 ②, are strongly exergonic and hence irreversible processes. As a consequence, decay of the ³Ru₇₇:STC population is expected to be mono-exponential. The TA reveals this is not the case, Fig. 6A pink circles. To account for the observed behavior several modifications and extensions of the reaction scheme were explored. In all cases the behavior was attributed to intramolecular electron transfer within Ru₇₇:STC monomers. There is good agreement between the dynamics of the charge separated states at 22 and 160 μM protein, Fig. S5A, and analytical ultracentrifugation, Fig. S2, defined the solution mass of Ru₇₇:STC at 22 μM as ≈15 400 Da, a value close that of 14 056 Da measured for the monomer by LC-MS.

As described fully in the Methods and Supporting Information, fits to different reaction schemes were obtained by solving a chemical Master equation assuming that all protein is initially ³Ru₇₇:STC. The only scheme to produce a reasonable fit to the ³Ru₇₇:STC decay included several kinetically distinct forms of the Ru-dye labelled protein. This is indicated in Scheme 2 where *x* indicates kinetically distinct populations *a*, *b* etc and *y* indicates the Ru-dye attachment site.



Scheme 2

Given the ultrafast time scale of charge separation we assign the kinetically distinct populations to different conformers of the Ru-labelled protein. This assumption is supported by MD simulations, described below, where different conformers do not interconvert on the timescale of charge separation and recombination. The multiple conformer model was then expanded to include reversible heme-heme electron transfer by replacing Ru₇₇⁺:STC⁻ of Scheme 2 with CS_x^{IV} and CS_x^{III} as illustrated in Scheme 3. In Scheme 3 it is assumed that the immediate product of charge separation contains reduced Heme IV, the site closest to the Ru-dye. This assumption is supported by calculations, as described in the following section. Electron migration along the heme chain is then included as the CS_x^{IV} ↔ CS_x^{III} interconversion where the electron resides on Heme III in CS^{III} and *k*_{III,IV} describes the rate constant for Heme IV → Heme III electron transfer. To keep the required fit parameters to a minimum, we assume that *k*_{III,IV} is the same for all conformers *x*. As we will see, Heme IV → Heme III electron transfer predominantly occurs in a single conformer that exhibits the slowest recombination kinetics, hence justifying the choice of a single rate constant *k*_{III,IV}.



Scheme 3

Scheme 3 produces an excellent description, evidenced by the lines in Figure 6, of the measured behavior when three conformers, *a* to *c*, participate with the individual contributions and rate constants presented in Table 1 bold. Without Heme IV \leftrightarrow Heme III electron transfer, the fit of the $\text{Ru}_{77}^+:\text{STC}^-$ population predicted by Scheme 2 at long-times (10 – 50 ns) is worse (see Fig. S6B, $R^2 = 0.966$ versus 0.971 without and with interheme electron transfer respectively). We also note that the fit presented in Fig. 6 with parameters in Table 1 includes the minimum number of kinetically distinct conformers needed to explain the TA data, and gives the best fit to the most certain experimentally defined population, that of $\text{Ru}_{77}^+:\text{STC}^-$, due to highest signal to noise. Models with more than three reactive conformers do not reproduce the data significantly better. There was no advantage to including electron transfer further along the STC heme chain, e.g., from Heme III \rightarrow Heme II, in the kinetic model. However, we do not rule out such electron transfer and return to consider this possibility further in the Discussion.

From the description of the $\text{Ru}_{77}:\text{STC}$ photochemistry presented in Figure 6 it is apparent that in addition to Heme IV \rightarrow Heme III electron transfer with $k_{\text{III,IV}} = 87 \times 10^6 \text{ s}^{-1}$, contributions from multiple conformers are required to account for the complete photocycle that extends from 0.2 pico- to 1 milli-second after excitation of the photosensitizer. For conformer *a* where charge separation and recombination are fastest and have rate constants much greater than $k_{\text{III,IV}}$, the model predicts $\text{Ru}_{77}^+:\text{STC}^-$ existing only as CS^{IV} , Fig. 6B red line. By contrast, in conformer *c* where charge separation and recombination are slowest, Heme IV \rightarrow Heme III electron transfer is competitive with charge recombination and a significant amount of CS^{III} is predicted in addition to CS^{IV} , Fig. 6B blue shaded and blue open respectively. Further consideration of the Heme IV \leftrightarrow Heme III electron transfer rate constants and their implications for the electron transfer mechanism is presented for the Discussion that concludes this paper.

Table 1: Electron transfer rate constants¹ for the Ru₇₇:STC photocycle. Values are derived from fit to the observed transients (bold) and from MD/DFT calculation as described in the text and Supporting Information.

		conformer		
		<i>a</i>	<i>b</i>	<i>c</i>
contribution (%)	Fit	22	33	45
k_{CS} (/10 ⁶ s ⁻¹)	Fit	41 700	3 300	261
	MD/DFT ³	333 000	66 700	256
k_{CR} (/10 ⁶ s ⁻¹)	Fit	30 300	11 400	1 230
	MD/DFT ³	7 190	6 710	588
$k_{III,IV}$ (/10 ⁶ s ⁻¹)	Fit ²	87		
	MD/DFT ^{2,4}	17		

¹ k_{CS} , k_{CR} and $k_{III,IV}$ are as defined in Scheme 3 and the parameters used for MD/DFT calculations are summarized in Tables S5 and S6.

² Assumed to be the same for all conformers.

³ Values for *a* averaged over two conformers (**1** and **2**) with shortest bpy-to-heme edge distances, see text for details.

⁴ From Jiang *et al* Ref ¹⁸, Table S4, state “(O,d)”.

Atomistic Description of Ru₇₇:STC Photochemistry. As noted above, a key feature of the kinetic model describing the Ru₇₇:STC photocycle is the presence of three Ru-conformers with different rate constants for electron exchange between the Ru-dye and Heme IV. We assessed the validity of this model through molecular docking and MD simulation to consider the nature of the predicted conformers. A structural model of Ru₇₇:STC was obtained by docking [Ru(II)(bpy)₂(4-CH₂-4'-methylbipyridine)]²⁺ to the crystal structure¹⁶ of STC (pdb identifier: 1M1Q) with serine 77 modified to cysteine (see *Methods* and Supporting Information for details). Of the 100,000 docking structures generated four were particularly low in energy and selected as initial structures for MD runs (total length 400 ns). Analysis of the MD trajectories shows that the label adopts four distinct conformations, see Fig. 7A where, for clarity, only three are shown. These conformations are stable on the nanosecond time scale and populate distinct low-energy basins characterized by dihedral angles τ_2 and τ_3 , Fig. 7B. In two conformers, **1** and **2**, a bpy ligand of the Ru-dye approaches the Heme IV ring very closely, bpy-to-heme edge-edge distance = 5.2 Å (thermal average), whereas in the other two conformers **3** and **4** the distances are larger, 6.1 and 8.5 Å, Fig. 7C. The corresponding distances to the next-nearest heme III are larger, by 3 Å or

more, suggesting that simultaneous electron injection into heme III is significantly slower and can be neglected to a good approximation.

For all four conformers we computed parameters relevant to charge separation and recombination of Scheme 3 using MD simulation and DFT electronic structure calculations as described in *Methods* and Supporting Information. The relevant frontier orbitals for charge separation from $^3\text{Ru}_{77}:\text{STC}$ to Heme IV are shown in Fig. 7D and the ones for charge recombination from $\text{Ru}_{77}^{\pm}:\text{STC}^{\mp}$ in Fig. S12. We find that electronic coupling between these states for charge separation is lower by approximately one order of magnitude for the conformer with the largest bpy-to-heme edge distance (**4**, Fig. 7E blue) compared to the conformers with the shorter distances (**1** and **2**, Fig. 7E black and red). A similar trend is obtained for the coupling between frontier orbitals for charge recombination. By contrast, the reorganization free energy, suitably corrected¹⁷ for non-ergodic effects on the present ultrafast timescale, is higher by about 0.2 eV for **4** than for **1** and **2**. Using these electron transfer parameters three distinct sets of rate constants are calculated, all spanning the ps-ns regime (summarized in Table 1). One set arises from the two conformers with the smallest distances, **1** and **2**, and one set each from **3** and **4**. These sets are assigned to the kinetically distinct conformers *a* (**1+2**), *b* (**3**) and *c* (**4**) proposed in our model of the TA data in order of decreasing charge separation rate constants, see Table 1. The sets of charge separation and charge recombination rate constants, obtained here from DFT and MD calculations without empirical adjustments, are in good agreement with predictions from empirical tunneling models⁴⁹⁻⁵⁰ as described in Table S6. We note that additional reactive conformers may exist in addition to the ones we have found, but we would expect their electron transfer kinetics to be similar to those of either *a*, *b* or *c*.

Comparison of measured and calculated rates for charge separation and recombination, Table 1, shows they are in reasonable agreement, typically within an order of magnitude, and gives confidence in our kinetic model for the photochemistry of $\text{Ru}_{77}:\text{STC}$, Scheme 3. A better agreement between computation and experiment can hardly be expected given that model structures of the Ru-labeled STC are used for the calculations and that charge separation and charge recombination rates are very sensitive to small changes in donor/acceptor distance and orientation. We note that the high rate constants for charge separation are consistent with a quantum yield for charge separation of ≈ 1 . The rates of charge separation and recombination are also seen to be higher than for many other studies of light-driven protein electron transfer, where the relevant redox centers are separated by larger distances and electron transfer slower than for $\text{Ru}_{77}:\text{STC}$.^{22,24-27} This can in part explain why to date we have been unable to employ flash-quench approaches effectively with this system.

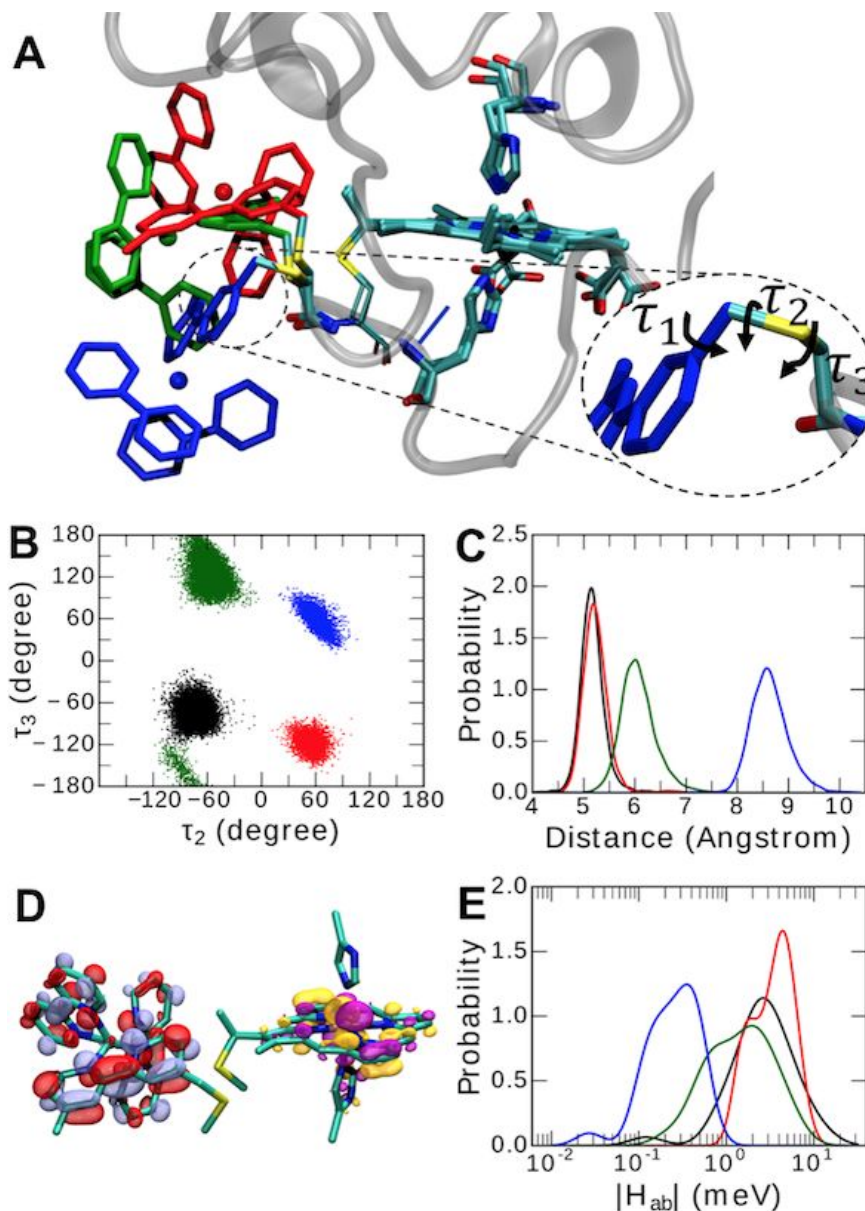
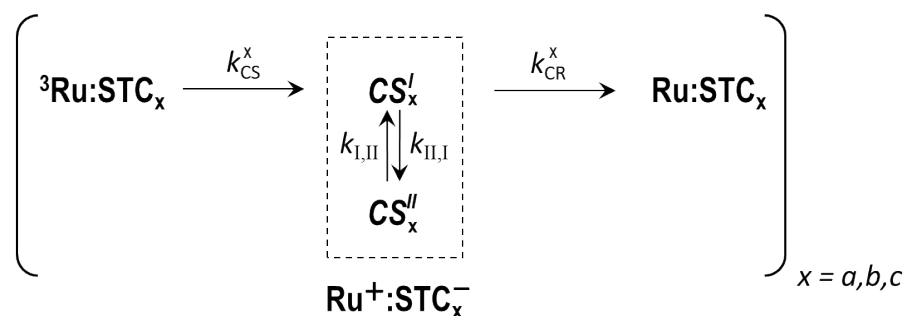


Figure 7. Molecular models of the reactive conformers of Ru₇₇:STC.

(A) Three of the four low-energy conformers obtained from molecular docking. The three structures are aligned with respect to the Fe atom of Heme IV, shown in stick representation (C cyan, N blue, O red and S yellow). The [Ru(bpy)₃]²⁺ label of conformers **1**, **3** and **4** are shown in red, green and blue, respectively, and the protein backbone is depicted in grey. The Ru-label of the lowest energy conformer, **2**, adopts a similar conformation as **1** but is not shown to enhance clarity of the illustration. (B) Scatter of dihedral angles τ_2 and τ_3 (defined in (A) inset), as obtained from MD simulation. (C) Probability distributions of the smallest distance between the C atoms of the bpy ligands and the Heme IV edge. (D) Pair of redox active frontier orbitals contributing to electronic coupling for charge separation. (E) Probability distribution of electronic coupling for ${}^3\text{Ru}_{77}:\text{STC} \rightarrow \text{Ru}_{77}^+:\text{STC}^-$ (CS^{IV}) charge separation as obtained from MD simulations and DFT calculations on the conformers, see main text and SI for details. The colour code in (B), (C) and (E) is the same as in (A) with the data for conformer **2** in black. In Table 1, **1** and **2** are assigned to conformer *a*, and conformers **3** and **4** to *b* and *c*, respectively.

Photochemistry and Electron Transfer Dynamics of Ru₂₃:STC and Ru₁₀:STC. In view of the conclusions reached from experiment and calculation of the properties of Ru₇₇:STC, a similar approach was used to investigate the consequences of light-driven electron transfer into STC Heme I, at the opposite end of the heme chain to Heme IV, Fig. 1. Ru₁₀:STC and Ru₂₃:STC were used for these studies. The Ru(II)-dye photoluminescence is very significantly quenched⁹ (>98%) when attached to both proteins which are monomeric in the conditions of the corresponding experiments. The dynamics of the corresponding charge separated states are in good agreement at ≈ 20 and $160 \mu\text{M}$ protein and analytical gel filtration produced elution profiles indistinguishable from that of monomeric Ru₇₇:STC.

The transient populations of Ru₂₃:STC after excitation into the MLCT-band of the Ru(II)-dye are presented in Fig. 8 circles. Noting that decay of ³Ru₂₃:STC is multi-exponential, multiple possible reaction schemes were again explored to describe the intramolecular electron transfer dynamics. Scheme 4 ($\gamma = 23$) describes three reactive conformers (*a-c*) and electron migration along the heme chain, now as $CS_x^I \leftrightarrow CS_x^{II}$ interconversion in the charge separated state. This Scheme with the parameters of Table 2 (bold) includes the minimum number of kinetically distinct conformers needed to explain the TA data, Fig. 8 lines, and gives the best fit to the most certain experimentally defined population, that of Ru₂₃⁺:STC⁻. The slower reacting conformer (*c*) displays significant Heme I \rightarrow Heme II electron transfer with a rate constant $125 \times 10^6 \text{ s}^{-1}$. Without interheme electron transfer the fit of the Ru₂₃⁺:STC⁻ population, Scheme 2, at long-times (3 – 500 ns) is worse (see Fig. S7, $R^2 = 0.928$ versus 0.972 without and with interheme electron transfer respectively). Models with more than three reactive conformers do not reproduce the data significantly better and there was no advantage to including Heme II \rightarrow Heme III electron transfer in the model.



Scheme 4

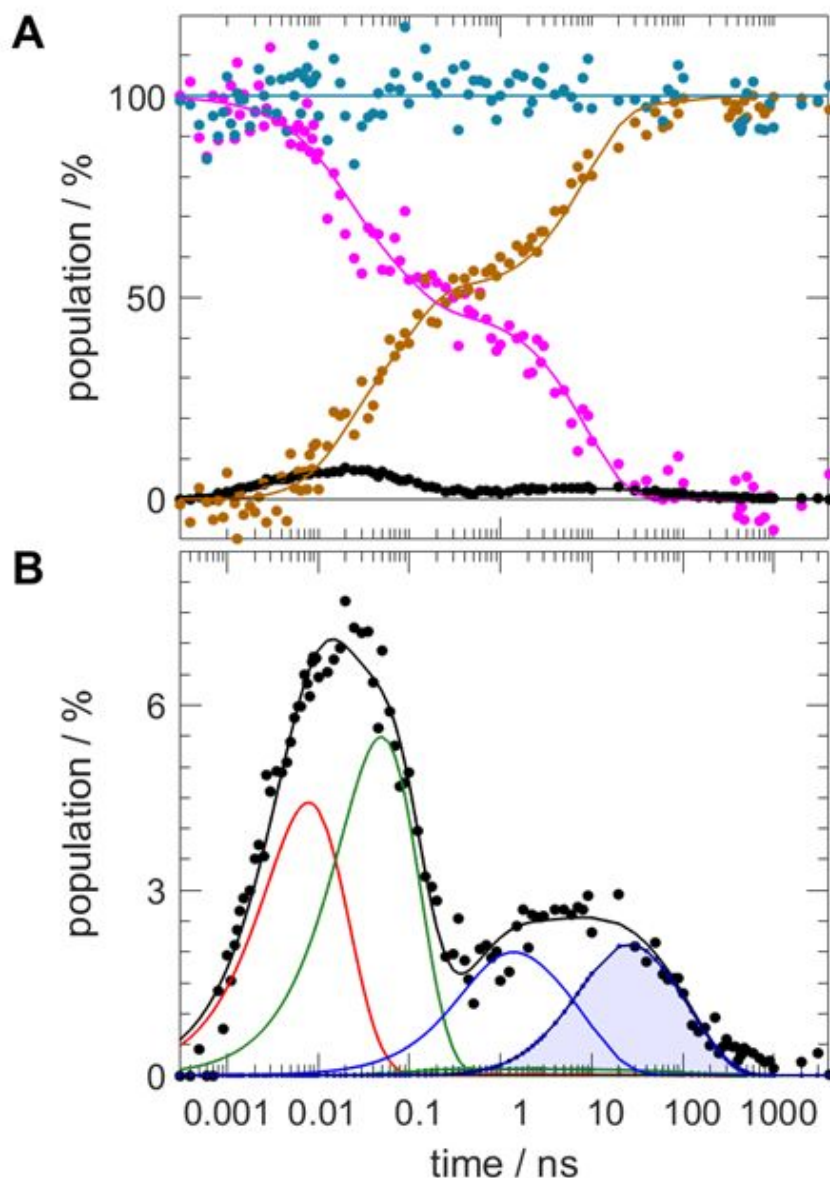


Figure 8. Progress of the Ru₂₃:STC Photocycle

(A) Evolution of the experimentally defined concentrations (circles) of ³Ru₂₃:STC (pink) and Ru₂₃⁺:STC⁻ (black) together with recovery of Ru₂₃:STC (brown) and the total concentration of these species (dark cyan). Fit to Scheme 4 (lines) with the parameters of Table 2 (bold). The population of ³Ru₂₃:STC at 0.7 ps was defined as 100%. (B) Expanded view of the population of Ru₂₃⁺:STC⁻ shown in A) from experiment (black circles) and fit to Scheme 4 (black line) with the Table 2 parameters (bold). Contributions to the fit from CS^I (lines, no fill) and CS^{II} (lines, shaded fill) are shown for conformers *a* (red), *b* (green) and *c* (blue). Note, contributions to CS^{II} from conformers *a* and *b* are very small and hardly visible.

Table 2: Electron transfer rate constants¹ for the Ru₂₃:STC photocycle. Values are derived from fit to the observed transients (bold) and from MD/DFT calculation.

		conformer		
		<i>a</i>	<i>b</i>	<i>c</i>
contribution (%)	Fit	27	26	47
k_{CS} (/10⁶ s⁻¹)	Fit	62 500	11 900	113
k_{CR} (/10⁶ s⁻¹)	Fit	250 000	31 200	2 170
$k_{II,I}$ (/10⁶ s⁻¹)	Fit²	126		
	MD/DFT ^{2,3}	8.7		

¹ k_{CS} , k_{CR} and $k_{II,I}$ are as defined in Scheme 4.

² Assumed to be the same for all conformers.

³ From Jiang *et al* Ref¹⁸, Table S4, state “(O,d)”.

For Ru₁₀:STC, excitation into the dye MLCT-band again resolved spectral features characteristic of the species in Scheme 1 but two models produced equally good fits to the transient populations; Scheme 4 with three reacting conformers and heme-heme electron transfer, and Scheme 2 with four reacting conformers but no heme-to-heme electron transfer. The corresponding fits and parameters are summarized in Fig. S8 and Table S4 respectively. With four reacting conformers the variations in k_{CS}/k_{CR} are $> 10^3$ and, while it is difficult to place an upper limit on this ratio, we consider values exceeding 10^2 as unlikely. With three reactive conformers and heme-to-heme electron transfer in the slowest, all conformers have $k_{CS}/k_{CR} < 11$. For this model the predicted Heme I \rightarrow Heme II electron transfer rate constant of $143 \times 10^6 \text{ s}^{-1}$ is very similar to the value of $125 \times 10^6 \text{ s}^{-1}$ predicted from modelling the TA of Ru₂₃:STC.

Discussion:

Ultrafast TA has allowed us to measure with high temporal resolution and signal-to-noise the visible-light driven electron transfer dynamics in a multiheme cytochrome photosensitized by site-specific labelling with a Ru(II)(bpy)₃ dye. For all proteins, the quantum yields for charge separation were high (≈ 1) and the photochemistry well-described by kinetic models, supported by MD simulation and DFT calculation, that include heme-heme electron transfer in competition with charge recombination. Our results have two important implications. Firstly, they demonstrate that TA of Ru-labelled multiheme cytochromes provides a promising platform for determination of heme-to-heme electron transfer rates. These values are difficult to access experimentally due to the identical chemical nature of the heme groups, yet they are of major interest as they define the upper limit for electron flow in biological extracellular respiration¹¹ and possibly in multiheme cytochrome-based bionanoelectronic junctions^{21, 51-52} at humid/wet conditions. Secondly, as multiheme cytochromes have an ability to accumulate photoenergized electrons in the presence as sacrificial electron donors⁹, the results provide a framework for rational development of photosensitized multiheme cytochromes to assist the coupling of one-electron photochemistry and multi-electron chemical transformations for visible-light driven redox catalysis.

The heme-heme electron transfer rate constants defined here by TA for Ru-labeled STC and summarized in Figure 9 (blue) are in line with the predictions⁵³ from NMR studies of intramolecular electron exchange at rates $> 10^4 \text{ s}^{-1}$, and are consistent with both stopped-flow kinetic analysis⁵⁴ of STC reacting with non-physiological redox partners, and MD simulations⁵⁵ considering electron exchange between STC and the surface of hematite ($\alpha\text{-Fe}_2\text{O}_3$). Also presented in Figure 9 (black) are the rate constants for heme-to-heme electron transfer predicted previously¹⁸ for native (i.e. unlabelled) STC using similar DFT and MD methodology as in the present study. We find that the computed rate constants are about an order of magnitude smaller than the experimental values, a factor of 7 for Heme IV \rightarrow Heme III and a factor of 14 for Heme I \rightarrow Heme II electron transfer. A possible reason for the greater discrepancy for the latter is that the reorganization free energy, 1.08 eV, might be overestimated in the calculations. For comparison, the predicted value¹⁸ for Heme IV \rightarrow Heme III electron transfer is 0.88 eV. The error of these estimates due to inaccuracies of the protein and water force field is typically in the order of 0.1 eV.¹⁷ If the reorganization energy for Heme I \rightarrow Heme II electron transfer is lowered by 0.1 eV, to 0.98 eV, the deviation with respect to experiment decreases from a factor of 14 to 5. An additional source of the discrepancy could be the presence of the Ru-label in the experimental measurements, which could give rise to a slight reduction in the reorganization free energy for the adjacent Heme IV \rightarrow Heme III and Heme I \rightarrow Heme II electron transfer. The terminal hemes, in particular Heme I, have a very high solvent accessible surface area in unlabeled STC and the attachment of the bulky and hydrophobic Ru-label in their vicinity will lower solvent access to the terminal hemes and therefore, most likely, lower reorganization free energy and increase the electron transfer rate. We do

not think this effect is very large but possibly large enough to explain at least part of the remaining difference between computed and experimental rate constants.

Taking into account all of these uncertainties, the computed values give further support for the interpretation of the TA data in terms of light-initiated electron transfer across the T-shaped Heme IV/III and Heme I/II pairs. As reported in our previous computational work¹⁸, the mechanism of heme-heme electron transfer is electron hopping in the non-adiabatic regime assisted by weak mixing of the Fe-heme *d* orbitals and sulfur 3*p* orbitals. The sulfur orbitals are located in the cysteine linkages binding the hemes to the protein and inserting into the space between the hemes having a T-shaped configuration. If the Fe-S orbital mixing is excluded, the computed rate constants for Heme IV \leftrightarrow Heme III and Heme I \leftrightarrow Heme II electron transfer were predicted to be 18-fold and 54-fold lower, respectively, and the deviation with the experimental rates reported herein would increase correspondingly. Hence, the experimental rates determined in this work support our hypothesis that orbital mixing accelerates electron transfer between these two T-shaped heme pairs. This effect is not restricted to STC but indeed has also been predicted for T-shaped and co-planar heme pairs in the deca-heme cytochromes MtrC and MtrF.²¹

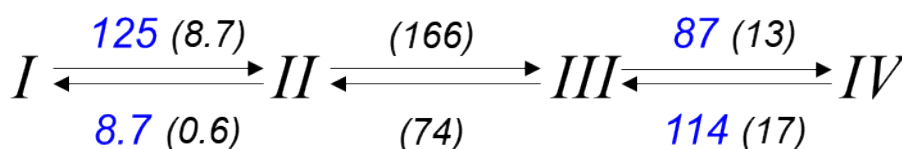


Figure 9. STC heme-heme electron transfer rate constants. Values ($\times 10^6 \text{ s}^{-1}$) measured by TA in this work (blue) and calculated previously¹⁸ for the 1e⁻ reduced protein (black).

In addition to their presence in STC, and the multiheme cytochromes MtrC⁵⁶ and MtrF⁵⁷ that participate in electron transfer across the outer membrane of *S. oneidensis*, T-shaped heme pairs are found⁷⁻⁸ alternating with stacked heme pairs in the electrically conductive extracellular *G. sulfurreducens* OmcS filaments that extend over distances $> 10 \mu\text{m}$. Our results suggest rate constants for electron transfer between these T-shaped hemes can be accurately predicted using the mechanisms and approaches described above. For STC those methods predict rate constants for electron transfer between the stacked Heme II/III pair that are approximately an order of magnitude greater than for the T-shaped heme pairs. This may explain why there is no benefit to including electron migration

further along the STC heme chain, e.g. Heme II \rightarrow Heme III after injection into Heme I, and Heme III \rightarrow Heme II after injection into Heme IV, in our kinetic models.

In closing we note that in a previous study²³ by Kokhan and colleagues reported ultra-fast TA of a tri-heme containing cytochrome, PpcA, photosensitized for light-driven electron transfer in the same way as the STC proteins described here. Charge separated states were generated from photoexcitation of the fully oxidized, and fully reduced, PpcA. Nevertheless the authors were unable to find evidence for heme-to-heme electron transfer. In view of the results presented here we suggest that this does not preclude light-driven heme-to-heme electron transfer in Ru:PpcA. Rather this process may have gone undetected due to contributions from a number of reactive conformers and/or the relative rates of electron transfer in the corresponding photocycles. As illustrated here the evidence for heme-to-heme electron transfer is compelling for Ru₇₇:STC and Ru₂₃:STC but less clear cut for Ru₁₀:STC. Moving forward experiments exploring the opportunities to obtain direct evidence for heme-heme electron transfer through pump-probe spectroscopies of multi-heme cytochromes containing spectroscopically distinct, e.g. His/Met coordinated or high-spin heme, at a defined location, are ongoing in our laboratories.

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Abbreviations: CR, charge recombination; CS, charge separation; DFT, density functional theory; LMCT, ligand to metal charge transfer; MD, molecular dynamics; MLCT, metal to ligand charge transfer; SHE, standard hydrogen electrode; STC, small tetraheme cytochrome (CctA) from *S. oneidensis* MR-1; TA, transient absorbance.

Supporting Information: Full methods of protein preparation, data collection, analysis, fitting to kinetic models, molecular docking, MD simulations, calculation of electron transfer parameters and rate constants. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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TOC Graphic

